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ORIGINAL ARTICLE

CINNAMIC ACID AND ITS DERIVATIVES IN THE HERBAL MIXTURES AND THEIR ANTIDIABETIC ACTIVITY

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Abstract

Rezumat

Datorită gamei largi de substanțe biologic active, amestecurile de produse vegetale pot influența dezvoltarea diabetului zaharat și complicațiile acestuia. Acidul cinamic și derivații săi au proprietăți antidiabetice și antioxidante. S-au investigat cinci eșantioane de amestecuri de plante cu un conținut potențial ridicat al acestor compuși. Folosind analiza HPLC s-au evaluat cantitativ 4 derivați ai acidului cinamic, și anume acizii clorogenic, cafeic, ferulic și sinapic din proba 1 a amestecului de plante; acid cinamic și 4 derivații ai săi, acizii clorogenic, p-cumaric, ferulic și sinapic, în probele 2 - 5. Rezultatele noastre au arătat că acidul fenolic predominant a fost acidul clorogenic, cu un conținut de 149,8 μg/g în prima probă, 180,1 μg/g în a doua probă, 306,7 μg/g în proba 3, 182,7 μg/g în proba 4 și 257,57 μg/g în proba 5. În timpul testării activității antidiabetice *in vitro* am evaluat potențialul inhibitor al produselor vegetale asupra α-amilazei și α-glucozidazei. Rezultatele studiului au arătat că amestecurile de produse vegetale studiate au potențial antidiabetic.

Keywords: diabetes mellitus, vegetal products, cinnamic acid, chlorogenic acid, p-coumaric acid, ferulic acid, sinapic acid

Introduction

Diabetes mellitus is one of the most important health issues worldwide, which requires immediate solutions, as the epidemiological situation is alarming – the number of patients is growing rapidly each year, leading to increased disability and mortality due to the development of macro- and microangiopathies [3]. According to the official data of the International Diabetes Federation (2019), the incidence of diabetes in the world is projected to increase 1.5 times by 2030, amounting to more than 500 thousand patients [11]. An important problem of pharmacovigilance is that existing pharmacotherapy can effectively reduce hyperglycaemia, but it is not always able to stabilize fluctuations in glycaemic values during the day and maintain it at an optimal level. This leads to the formation of a cascade of pathological processes excessive glycation and inactivation of the body's antioxidant defence system, triggering the processes

of free radical oxidation of lipids and, as a consequence, the development of oxidative stress, which leads to the development and progression of diabetic angiopathies [3, 23]. Therefore, the optimization of existing antidiabetic pharmacotherapy, research and development of new drugs for the prevention and treatment of this disease and its complications are currently very important issues in modern pharmacy and medicine. One of these areas is using herbal remedies, either as monotherapy for the prevention or in the mild stages of the disease or in the combination with traditional therapy in more severe forms of the disease [13, 15, 17]. Phytotherapy is a justified method for the prevention and treatment because it has some advantages, such as relatively low toxicity, mild pharmacological effects and possibility to be used for long periods without significant side-effects, and it often well combines with synthetic drugs [8, 17, 20].

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Particular attention should be paid to the combinations of different medicinal plants, because such herbal mixtures will have more biologically active substances (phytocomplexes) that may influence many aspects of the pathogenic mechanism of diabetes mellitus development and its complications [26, 27]. In this regard, some important biologically active substances are the polyphenols [9]. They comprise a wide range of phytochemical compounds from the following groups: phenolic acids, flavonoids, tannins, stilbenes, coumarins, lignans. Phenolic acids are usually classified into two major groups: benzoic acids, containing seven carbon atoms (C6-C1), and cinnamic acids, comprising nine carbon atoms (C6-C3) [33]. Among various biological activities, cinnamic acid and its derivatives are associated with the ameliorating effects on diabetes and its complications, mainly due to the presence of hypoglycaemic and antioxidant activities [1, 12]. Their antidiabetic effect is implemented by different mechanisms of action, including stimulation of insulin secretion, improvement of pancreatic β-cell functionality, inhibition of gluconeogenesis, intensification of glucose uptake, delay of carbohydrate digestion and glucose absorption, inhibition of protein glycation and insulin fibrillation. The antioxidant activity of these compounds is due to the fact that cinnamic acid and its derivatives

neutralize free radicals by cleavage of the hydrogen atom [1, 2, 12, 18].

Thus, for this purpose, it becomes useful to study the phytochemical compounds, namely the cinnamic acid and its derivatives from the group of phenolic acids within the herbal mixtures with previously *in vivo* antidiabetic activity investigated [25].

Materials and Methods

Plant materials

The herbal raw materials harvested from June to August 2019 in the Ternopil region (Ukraine) were used. After harvesting, the raw materials were dried, crushed and stored according to the general GACP requirements [36]. The plants were identified in the Department of Pharmacognosy with Medical Botany, I. Ya. Horbachevsky Ternopil National Medical University, Ternopil, Ukraine. The voucher specimens of herbal raw materials have been deposited in the departmental herbarium for future records. For the study, five different herbal mixtures with reliable antidiabetic activity established during *in vivo* pharmacological studies [25] were used. The composition of the mixtures is given in Table I.

 $\begin{tabular}{ll} \textbf{Table I} \\ \textbf{Composition of the herbal mixtures} \\ \end{tabular}$

Herbal mixtures	Herbal drug component	Percentage in the mixture (%)	Relative ratio
	Urtica dioica leaf	26.32	5
Sample 1	Cichorium intybus roots	26.32	5
	Rosa majalis fruits	21.05	4
	Elymus repens rhizome	15.79	3
	Taraxacum officinale roots	10.52	2
	Arctium lappa roots	26.32	5
	Elymus repens rhizome	26.32	5
Sample 2	Zea mays columns with stigmas	21.05	4
	Helichrysum arenarium flowers	15.79	3
	Rosa majalis fruits	10.52	2
	Inula helenium rhizome with roots	10.0	1
	Helichrysi arenarium flowers	20.0	2
Cample 2	Zea mays columns with stigmas	20.0	2
Sample 3	Origanum vulgari herb	20.0	2
	Rosa majalis fruits	20.0	2
	Taraxacum officinale roots	10.0	1
	Cichorium intybus roots	26.32	5
	Elymus repens rhizome	26.32	5
Sample 4	Helichrysum arenarium flowers	21.05	4
	Rosa majalis fruits	15.79	3
	Zea mays columns with stigmas	10.52	2
	Urtica dioica leaf	20.0	1
	Taraxacum officinale roots	20.0	1
Sample 5	Vaccinium myrtillus leaf	20.0	1
_	Rosa majalis fruits	20.0	1
	Mentha piperita herb	20.0	1

Chemicals and standards

Chemical reference substances (CRS) of chlorogenic acid, caffeic acid, *p*-coumaric acid, ferulic acid, sinapic acid, cinnamic acid, acarbose were of primary reference

standard grade (\geq 95% purity HPLC) and were purchased from Sigma-Aldrich Chemical Company (Germany), as well as α -amylase, α -glucosidase. Water used in

the studies was produced by MilliQ Gradient water deionization system (USA).

Extraction of cinnamic acid and its derivatives for HPLC assay

The samples of herbal raw materials were ground into a powder by laboratory mill, then about 500 mg (accurately weighed) was selected and placed into vial with 5 - 10 mL of 60% methanol. The extractions were carried out in an ultrasonic water bath at 80° C for 4 hours. The resulting extracts were centrifuged at 3000 rpm and filtered through disposable membrane filters with pores of $0.22 \ \mu m \ [30]$.

Instrumentation and conditions of HPLC analysis The content of cinnamic acid and its derivatives in the samples of the herbal mixtures was studied by high performance liquid chromatography using 3D LC System from Agilent Technologies 1200 (USA) [22]. The separation was performed on a Zorbax SB-Aq chromatographic column (4.6 mm \pm 150 mm, 3.5 μm) (Agilent Technologies, USA) with thermostat temperature 30°C, injection volume of samples 4 µL and flow rate 0.5 mL/min at the gradient elution with the mobile phases - methanol (A) and 0.1% solution of formic acid in water (B). Elution was performed in a gradient mode: 0 min - A (25%):B (75%); 25 min – A (75%):B (25%); 27 min – A (100%):B (0%); 35 min - A (100%):B (0%). The registration of signal was done at 250 nm and 275 nm and fixation of absorption spectra in the range of 210 - 700 nm.

To identify the components, the obtained spectra were analysed by comparing the retention times (t_R) of CRS of cinnamic acid and its derivatives. Quantitative analyses were performed using the peaks areas.

Preparation of extracts

The samples of herbal raw materials (10 g) were placed into a 100 mL conical flask with 120 mL distilled water. The extractions were carried out in a water bath for 30 min. The resulting extracts were filtered using Whatmann filter paper no1. Then the filtrates were evaporated by rotary evaporator and were lyophilized to dryness. The lyophilized powders of each herbal mixture were stored at 4°C for further use.

α-amylase inhibition

The method is based on enzyme inhibition, so the transformation of starch to reducing oligosaccharides that react with 3,5-dinitrosalicylic acid is blocked. A total of 500 μ L of samples of the studied extracts with a range of concentrations between 100 and 1000 μg/mL were added to 500 μL of 0.20 mM phosphate buffer (pH 6.9 with 0.006 M sodium chloride) containing α-amylase solution (0.5 mg/mL) and were incubated at 25°C for 10 min. Thereafter, it was added 500 µL of (1% w/v) starch solution in 0.02 M sodium phosphate buffer (pH 6.9 with 0.006 M sodium chloride) to each tube and was incubated at 25°C for 10 min. The reaction was stopped with 1.0 mL of 3,5-dinitrosalicylic acid colour reagent (12.0 g of sodium potassium tartrate tetrahydrate in 8 mL of 2 M NaOH and 96 mM 3,5-dinitrosalicylic acid solution). Then the tubes were incubated in the boiling water bath for 5 min and cooled to room temperature. The reaction mixture was diluted by adding 10 mL of distilled water and absorbance was measured at 540 nm using the spectrophotometer Shimadzu 1800-UV (Japan). Experiments were performed in triplicate. Acarbose was used as a positive control [21].

α-glucosidase inhibition

The method is based on the inhibition of α -glucosidase that catalyses the hydrolysis of p-nitrophenyl-D-glucopyranoside to p-nitrophenol. A total of 20 μ L of samples of the studied extracts with a range of concentrations between 100 and 1000 μ g/mL were added to 50 μ L potassium phosphate buffer 0.1 M (pH 6.8) and 10 μ L α -glucosidase 0.25 U/mL and were incubated at 37°C for 10 min. Then 10 μ L 5 mM of p-nitrophenyl- α -D-glucopyranoside was added and further incubated for 30 min. the reaction was stopped with 50 μ L of Na₂CO₃ 0.1 M. The absorbance was measured at 405 nm using the spectrophotometer Shimadzu 1800-UV (Japan). Experiments were performed in triplicate. Acarbose was used as a positive control [14].

Calculation of 50% Inhibitory Concentration (IC50) The inhibitory concentration of the water extracts of the herbal mixtures required to inhibit the activity of the enzyme by 50%, IC50 was calculated by regression analysis using the percentage scavenging activities at five different concentrations of the extracts. Inhibition (I %) was calculated using the following equation:

% Inhibition = $\frac{\text{Absorbance of control} - \text{Absorbance of sample}}{\text{Absorbance of control}} \times 100$

Results and Discussion

The evaluation of cinnamic acid and its derivatives in the herbal mixtures

The results of qualitative and quantitative analyses of cinnamic acid and its derivatives in the herbal mixtures are represented in Table II.

During HPLC analysis were identified 4 derivatives of cinnamic acid, such as chlorogenic, caffeic, ferulic

and sinapic acids in the sample 1 of herbal mixture; cinnamic acid and 4 its derivatives, such as chlorogenic, *p*-coumaric, ferulic and sinapic acid in the samples 2 - 5 (Table II).

The in-vitro antidiabetic activity

The experimental studies regarding the *in vitro* antidiabetic activity of the investigated herbal mixtures in a concentrations range of $100 - 1000 \,\mu\text{g/mL}$ were performed by inhibition of α -amylase and α -glucosidase activity, using with acarbose as reference (Table III

and Table IV).

No. t _R , min		Identified substance -	Active principles (μg/g)				
			Sample 1	Sample 2	Sample 3	Sample 4	Sample 5
1.	10.18	chlorogenic acid	149.8 ± 0.28	180.1 ± 0.37	306.7 ± 0.56	182.7 ± 0.24	257.5 ± 0.29
2.	10.89	caffeic acid	18.9 ± 0.11	-	-	-	-
3.	13.91	p-coumaric acid	-	116.2 ± 0.5	92.5 ± 0.24	135.7 ± 0.27	26.1 ± 0.22
4.	15.16	ferulic acid	51.6 ± 0.21	12.5 ± 0.15	15.9 ± 0.18	31.5 ± 0.16	50.8 ± 0.19
5.	15.69	sinapic acid	28.5 ± 0.15	36.2 ± 0.14	33.2 ± 0.18	55.5 ± 0.19	25.1 ± 0.13
6.	18.14	cinnamic acid	-	58.8 ± 0.17	64.2 ± 0.18	75.7 ± 0.17	4.5 ± 0.11

Values are expressed as mean \pm SD (n = 5)

Table III α -amylase inhibition of water extracts of the samples of the herbal mixtures

Herbal mixtures	Concentration, µg/mL	Inhibition, %	IC50, μg/mL
	100	22.17 ± 3.65	, -
	200	30.97 ± 2.98	
Sample 1	400	41.15 ± 3.51	699.49
	800	52.97 ± 4.28	
	1000	61.64 ± 3.49	
	100	20.47 ± 3.31	
	200	28.58 ± 4.42	
Sample 2	400	39.13 ± 5.28	758.15
	800	51.27 ± 3.63	
	1000	59.34 ± 3.75	
	100	21.58 ± 3.53	_
	200	32.11 ± 3.94	
Sample 3	400	38.28 ± 2.37	781.76
	800	50.56 ± 3.63	
	1000	60.18 ± 3.74	
	100	20.65 ± 3.62	
	200	31.07 ± 2.86	
Sample 4	400	40.95 ± 3.73	700.17
	800	53.01 ± 3.85	
	1000	58.75 ± 3.92	
	100	23.04 ± 3.76	_
	200	31.65 ± 4.93	
Sample 5	400	42.82 ± 2.71	646.52
	800	54.47 ± 3.83	
	1000	63.18 ± 3.17	
	100	33.98 ± 1.92	
	200	47.37 ± 2.13	
Acarbose (reference)	400	58.75 ± 2.46	246.22
	800	69.58 ± 2.06	
	1000	75.94 ± 1.99	
maan + CD (n = 2)			

Values are expressed as $mean \pm SD (n = 3)$

Herbal mixtures	Concentration, µg/mL	Inhibition, %	IC50, μg/mL
	100	30.69 ± 3.12	
	200	41.16 ± 3.17	
Sample 1	400	54.95 ± 2.98	328.16
	800	62.10 ± 3.08	
	1000	73.86 ± 3.05	
•	100	29.49 ± 2.92	
	200	39.58 ± 3.18	
Sample 2	400	52.64 ± 3.14	359.57
	800	60.86 ± 3.03	
	1000	71.46 ± 3.28	

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Herbal mixtures	Concentration, µg/mL	Inhibition, %	IC50, μg/mL
	100	28.94 ± 3.09	_
	200	40.48 ± 2.67	
Sample 3	400	53.82 ± 2.75	342.73
	800	60.69 ± 3.04	
	1000	70.93 ± 3.15	
	100	29.83 ± 3.04	
	200	38.94 ± 3.15	
Sample 4	400	51.99 ± 3.03	369.50
	800	64.04 ± 2.61	
	1000	70.96 ± 2.19	
	100	32.85 ± 3.08	_
	200	43.91 ± 3.24	
Sample 5	400	57.07 ± 3.31	292.55
	800	65.19 ± 2.98	
	1000	75.38 ± 2.82	
	100	42.19 ± 2.12	
	200	49.84 ± 1.98	
Acarbose (reference)	400	62.11 ± 1.59	202.62
	800	74.09 ± 2.37	
	1000	89.73 ± 2.65	

Values are expressed as $\overline{\text{mean} \pm \text{SD} (\text{n} = 3)}$

The quantitative determination of cinnamic acid and its derivatives showed that the predominant phenylpropanoid was chlorogenic acid, its content was 149.8 $\mu g/g$ in the sample 1; 180.1 $\mu g/g$ in the sample 2; 306.7 μ g/g in the sample 3; 182.7 μ g/g in the sample 4; $257.57 \mu g/g$ in the sample 5 (Table II). The high content of chlorogenic acid in the studied mixtures is of interest, because it has a pronounced hypoglycaemic activity, due to the fact that it increases the use of glucose by skeletal muscles, improves glucose tolerance, stimulates insulin secretion by pancreatic β-cells. In addition, chlorogenic acid has the ability to potentiate the action of insulin, similar to the therapeutic action of metformin - synthetic antidiabetic drug of the biguanide class [19]. It is also important that chlorogenic acid has the ability to regulate lipid metabolism by lowering triglycerides, low-density lipoproteins and cholesterol, which is signify to prevent the development of cardiovascular diseases and microcirculatory complications - diabetic nephropathy, neuropathy and retinopathy, the formation of diabetic foot [24]. Diabetes mellitus is often accompanied by the development of metabolic syndrome, which is characterized by obesity, and chlorogenic acid helps to lose weight and reduce the accumulation of visceral fat, activates fat metabolism in the liver and helps to reduce plasma leptin level [24, 32]. In addition, chlorogenic acid has strong antioxidant properties, which is important for the prevention of diabetic angiopathies, the pathogenesis of which are activation of lipid peroxidation, inactivation of antioxidant protection system and development of oxidative stress. Antioxidant properties of chlorogenic acid are realized by cleavage of hydrogen atoms, which reduces the amount of free radicals, lipid peroxidation products and inhibits the development of oxidative stress [4]. The development of inflammatory

processes in patients with diabetes mellitus is an important problem that requires some adjustments to the pharmacotherapy. This problem may be solved by using chlorogenic acid that exhibits anti-inflammatory properties, which decreases the oedema, suppresses the pro-inflammatory cytokines, reduces the neutrophil infiltration [34].

During HPLC analysis it was established the content of p-coumaric acid in four samples of herbal mixtures and it was 116.2 μ g/g in the second sample, 92.5 μ g/g in sample 3, 135.7 μ g/g in sample 4 and 26.1 μ g/g in sample 5 (Table II). p-Coumaric acid has many biological activities, such as hypoglycaemic, important antioxidant, anti-inflammatory, hepato-renal protective effects and protects the pancreas from free radical damage. Thus, p-coumaric acid is a good agent for lowering blood glucose, reducing oxidative stress, improving antioxidant status, reducing inflammation [10, 31]. Our study showed that the content of ferulic acid was 51.6 μ g/g in sample 1, 12.5 μ g/g in sample 2, 15.9 μ g/g in sample 3, 31.5 μ g/g in sample 4 and 50.8 μ g/g in sample 5 (Table II). Ferulic acid is a very important agent for the prevention and treatment of diabetes because it has hypoglycaemic effect, which is manifested by various mechanisms – inhibition of α -glucosidase, stimulation of insulin secretion, increased glucose utilization; antioxidant effect due to its ability to prevent protein glycation and lipid peroxidation of the membrane, neutralizing the formed free radicals

The results of our research showed that the content of sinapic acid was $28.5 \,\mu\text{g/g}$ in sample $1,\,36.2 \,\mu\text{g/g}$ in sample $2,\,33.2 \,\mu\text{g/g}$ in sample $3,\,55.5 \,\mu\text{g/g}$ in sample 4 and $25.1 \,\mu\text{g/g}$ in sample 5 (Table II). Sinapic acid has numerous peripheral actions, such as antioxidant, anti-inflammatory, hypoglycaemic, cardioprotective,

hepatoprotective and nephroprotective. Its hypoglycaemic effect is exerted by reducing insulin resistance and plasma glucose concentration, increasing the expression of the GLUT-4 gene in the skeletal muscle and cell sensitivity to insulin. In addition, it has the ability to regulate lipid metabolism by lowering triglycerides and cholesterol [6, 37].

It was established that the content of cinnamic acid in the second sample was 58.8 $\mu g/g$, in sample 3-64.2 $\mu g/g$, in sample 4-75.7 $\mu g/g$, in sample 5-4.5 $\mu g/g$ (Table II). Cinnamic acid exhibits antidiabetic properties by a variety of mechanisms of action, including stimulation of insulin secretion, improving the functioning of $\beta\text{-cells}$ of the pancreas, inhibiting gluconeogenesis, increasing glucose uptake, improving glucose tolerance. In addition, it has antioxidant, hepatoprotective and anti-inflammatory effects [1, 16].

Caffeic acid was identified only in sample 1 and its content was 18.9 µg/g (Table II). It has high antioxidant, antidiabetic, anti-inflammatory effects and has the ability to protect the pancreas from destruction by free radicals and lipid peroxidation products [1, 29]. According to the literature, the vegetal products Helichrysum arenarium flowers, Urtica dioica leaf, Rosa majalis fruits, Zea mays columns with stigmas, Origanum vulgari herb, Vaccinium myrtillus leaf and Mentha piperita herb, contain cinnamic acid and its derivatives [2, 7, 9, 12, 17, 18]. This aspect was also which was confirmed in our study by for the vegetal products from the herbal mixtures studied by our group. The relationship between the increase in the inhibitory activity of α-amylase and α-glucosidase and the concentration of aqueous extracts of herbal mixtures was evaluated. During the study of inhibition of αamylase enzyme it was established that the IC50 of the water extracts of the first sample was 699.49 µg/mL, the second sample $2 - 758.15 \mu g/mL$, sample 3 -781.76 μ g/mL, sample 4 - 700.17 μ g/mL and sample $5 - 646.52 \mu g/mL$ (Table III). Our results showed that the concentration required for 50% inhibition (IC50) of α-glucosidase enzyme was 328.16 μg/mL for sample 1, 359.57 μg/mL for sample 2, $342.73 \mu g/mL$ for sample 3, $369.50 \mu g/mL$ for sample 4 and 292.55 $\mu g/mL$ for sample 5 (Table IV). The IC50 value of standard drug acarbose against αamylase and α-glucosidase was 246.22 µg/mL and 202.62 µg/mL, respectively.

Thus, the establishment of the high content of cinnamic acid and its derivatives in the herbal mixtures studied and the assay of their *in vitro* antidiabetic activity by α -amylase and α -glucosidase inhibition may indicate these herbal mixtures as promising options for the management of diabetes and its complications.

Conclusions

The assay of cinnamic acid and its derivatives in the five different herbal mixtures with reliable antidiabetic activity was carried out. There were detected 4 derivatives of cinnamic acid, such as chlorogenic, caffeic, ferulic and sinapic acid in sample 1 of the herbal mixture (*Urtica dioica* leaf, *Cichorium intybus* roots, *Rosa majalis* fruits, *Elymus repens* rhizome, *Taraxacum officinale* roots); cinnamic acid and 4 its derivatives, such as chlorogenic, p-coumaric, ferulic and sinapic acid in the samples 2 - 5. The high content of cinnamic acid and its derivatives contributed to the antidiabetic activity, which was confirmed trough *in vitro* of inhibition of α -amylase and the α -glucosidase.

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Conflict of interest

The authors declare no conflict of interest.

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