

# SYNTHESIS, CHARACTERIZATION AND CYTOTOXICITY ASSESSMENT OF NEW 4-BENZYL-1,3-OXAZOLE DERIVATIVES INCORPORATING 4-[(4-BROMOPHENYL)SULFONYL]PHENYL FRAGMENT

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## Abstract

Herein we present the design, synthesis, characterization, and cytotoxicity assessment of seven compounds derived from phenylalanine that incorporate a 4-[(4-bromophenyl)sulfonyl]phenyl fragment: four open-chain products (one *N*-acyl- $\alpha$ -amino acid, one *N*-acyl- $\alpha$ -amino acyl chloride, two *N*-acyl- $\alpha$ -amino ketones), and three five-membered heterocycles with two heteroatoms (O and N): one 2-aryl-4-benzyl-1,3-oxazol-5(4*H*)-one, and two 2-aryl-4-benzyl-1,3-oxazoles with *p*-tolyl, and *m*-xylyl substituent, respectively, in position 5. Structures of new derivatives were assigned by elemental analysis, NMR spectroscopy, and other spectral methods (FT-IR, MS, UV-Vis). Evaluation of the purity of the compounds was realized by reversed-phase high-performance liquid chromatography. *Daphnia magna* toxicological test was used to assess the cytotoxicity of new compounds.

## Rezumat

Studiul prezintă proiectarea, sinteza, caracterizarea și evaluarea citotoxicității a șapte compuși derivați de la fenilalanină, care încorporează un fragment 4-[(4-bromofenil)sulfonil]fenil: patru produși cu catenă deschisă (*N*-acil- $\alpha$ -aminoacid, o clorură de *N*-acil- $\alpha$ -aminoacil, două *N*-acil- $\alpha$ -aminocetone) și trei heterocicluri pentaatomice cu doi heteroatomi (O și N): o 2-aril-4-benzil-1,3-oxazol-5(4*H*)-onă și 2-aril-4-benzil-1,3-oxazoli cu substituentul *p*-tolil și respectiv, *m*-xilil, în poziția 5. Structurile noilor derivați au fost atribuite prin analiză elementală, spectroscopie RMN și alte metode spectrale (FT-IR, SM, UV-Viz). Evaluarea purității compușilor a fost realizată prin cromatografie de lichide de înaltă performanță cu fază inversă. Testul toxicologic *Daphnia magna* a fost utilizat pentru a se evalua citotoxicitatea noilor compuși.

**Keywords:** *N*-acyl- $\alpha$ -amino acid, 4-benzyl-1,3-oxazol-5(4*H*)-one, 4-benzyl-1,3-oxazole, cytotoxic effect

## Introduction

1,3-Oxazoles have increasing importance in heterocyclic chemistry and drew the researchers' attention due to their biological and medicinal applications. Though parent 1,3-oxazole does not occur naturally, numerous natural derivatives containing a 1,3-oxazole core have been isolated, especially from marine invertebrates and microorganisms, some of them exhibiting remarkable therapeutical effects [1-4]. Additionally, a series of synthetic pharmaceutical molecules bearing 1,3-oxazole scaffold were reported to have antimicrobial [5], anti-diabetic [6], analgesic, anti-inflammatory [7], anticancer activity [8], etc. 5-Hydroxy-1,3-oxazoles exist in their corresponding

keto tautomer: 1,3-oxazol-5(4*H*)-ones which present cytotoxic [9, 10], antiviral [11], plant growth regulating properties [12]. Some representants of *N*-acyl- $\alpha$ -amino acids class show biological activities, such as anti-hypertensive [13], anticancer [14], mucolytic [15], antianaemic [16], anti-ulcer effect [17], and are specific antidotes in acute intoxications [18, 19]. *N*-Acyl- $\alpha$ -amino-ketones display anti-inflammatory [20], antiviral [21, 22] and antithrombotic action [23]. Besides, is known that numerous diaryl sulfones are used in therapy for their properties [24-26]. In an attempt to find new effective drugs, diaryl sulfone pharmacophore has been incorporated into a large number of derivatives with potential biological value [9, 10, 12, 27-29].

Given the scientific relevance of compounds from these classes and our previous researches [9, 10, 12, 30], in this article, we report the design, synthesis, and characterization of new *N*-acyl- $\alpha$ -amino acids, 1,3-oxazol-5(4*H*)-ones, *N*-acyl- $\alpha$ -amino acyl chlorides, *N*-acyl- $\alpha$ -amino ketones, and 1,3-oxazoles analogues derived from phenylalanine containing a 4-[(4-bromophenyl)sulfonyl]phenyl substituent into their structure with the aim of obtaining new bioactive products and to explore their biological action. For this purpose, the cytotoxic effect of compounds was assessed on *Daphnia magna* crustacean. Amidst the various methods of screening, this assay is fast, reproducible, cost-efficient, simple and can predict biological activity [27].

## Materials and Methods

### Chemistry

**General:** Chemicals and reagents were acquired from common commercial suppliers. Dichloromethane was anhydridized over anhydrous calcium chloride. Uncorrected melting points were measured on a Boëtius apparatus. UV-Vis spectra were registered for methanolic solutions ( $\approx 0.025$  mM) on an Analytik Jena AG Specord 40 spectrophotometer. FT-IR spectra were recorded in KBr pellets on a Bruker Vertex 70 spectrometer, absorption peaks being described as very strong, vs; strong, s; medium, m; weak, w. NMR spectra were acquired on a Varian Gemini 300 BB instrument at 300 MHz for  $^1\text{H}$  and 75 MHz for  $^{13}\text{C}$ . Combined 2D spectra (COSY, HETCOR) were also recorded. DMSO- $d_6$  or  $\text{CDCl}_3$  as deuterated solvents were used. Chemical shifts,  $\delta$ , are in parts *per* million (ppm), relative to the reference standard tetramethylsilane (TMS) signal. Coupling constants, *J*, are expressed in hertz (Hz). In the  $^1\text{H}$ -NMR spectra, signal multiplicity was assigned as singlet (s), doublet (d), doublet of doublets (dd), doublet of doublets of doublets (ddd), triplet (t), multiplet (m); broad is abbreviated b. GC-EI-MS analysis was registered on a Fisons Instruments GC 8000 with an electron impact quadrupole, MD 800 mass spectrometer detector, and a fused-silica capillary column coated with poly(5% diphenyl – 95% dimethylsiloxane) (SLB-5ms, 30 m  $\times$  0.32 mm,  $d_f$  0.25  $\mu\text{m}$ ). Dichloromethane was used as solvent and the helium flow rate was 2 mL/min. RP-HPLC chromatograms were performed on a Beckman System Gold 126 liquid chromatograph, with a System Gold 166 UV-Vis detector, a non-polar chromatography column (LiChrosorb RP-18, 25 cm  $\times$  4 mm, 5  $\mu\text{m}$  particle size), and a Rheodyne injection system. New compounds' purity and retention time,  $t_R$ , in minutes (min) are indicated. Elemental analysis was carried out on a Costech ECS 4010 instrument.

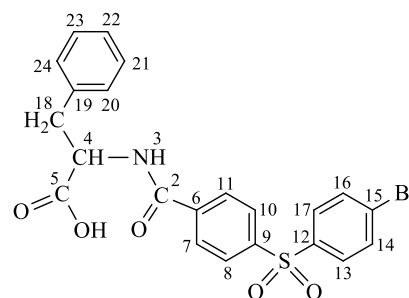
### Synthesis and characterization of compounds

#### Synthesis of 4-[(4-bromophenyl)sulfonyl]benzoyl chloride **2**

4-[(4-Bromophenyl)sulfonyl]benzoic acid **1** [31] (6.82 g, 20 mmol) was heated under reflux with an excess of thionyl dichloride (35 mL, 57.10 g, 480 mmol). Unreacted  $\text{SOCl}_2$  was distilled off the reaction mixture under *vacuum*. Obtained colourless crystals were used crude in the following reaction; yield = 99% (7.12 g), m.p. = 154 - 155°C (lit. [32] 154°C). FT-IR (KBr,  $\nu$   $\text{cm}^{-1}$ ): 3097s; 3040m; 1779vs; 1740vs; 1591m; 1572vs; 1470m; 1333vs; 1302s; 1287s; 1162vs; 888vs; 851s; 753vs; 730vs; 647vs; 575vs.

#### Synthesis of 2-{4-[(4-bromophenyl)sulfonyl]benz-amido}-3-phenylpropanoic acid **3**

Phenylalanine (3.30 g, 20 mmol) was dissolved in 1 N NaOH solution (20 mL). To this solution cooled to 0 - 5°C, a solution of raw 4-[(4-bromophenyl)sulfonyl]benzoyl chloride **2** (7.19 g, 20 mmol) in anhydrous dichloromethane (45 mL), and a 2 N NaOH solution (10 mL), respectively were added simultaneously, dropwise, under stirring, for 30 min. Reaction mixture stirring was continued for 1 h at room temperature. The separated aqueous layer was acidified with 2 N HCl. Precipitated solid was isolated by filtration, washed with water, dried, and recrystallized as white acicular crystals; yield = 92% (8.98 g); m.p. = 179 - 180°C (water).



**Figure 1.**

Atoms numbering of compound **3** used for the assignment of NMR signals

UV-Vis ( $\text{CH}_3\text{OH}$ ,  $\lambda$  nm) (lg  $\epsilon$ ): 202.6 (4.47); 250.9 (4.17).

FT-IR (KBr,  $\nu$   $\text{cm}^{-1}$ ): 3356s; 3085m; 3058m; 3027m; 2978m; 2938m; 2869m; 2703w; 2640w; 2590w; 2530w; 1734vs; 1622vs; 1574s; 1547vs; 1492m; 1456m; 1447m; 1323vs; 1310s; 1295vs; 1161vs; 852m; 621s; 570s.

$^1\text{H}$ -NMR (DMSO- $d_6$ ,  $\delta$  ppm, *J* Hz): 3.04 (dd, 13.7, 10.4, 1H, H-18); 3.20 (dd, 13.7, 4.4, 1H, H-18); 4.64 (m, 1H, H-4); 7.10-7.30 (m, 5H, H-20 – H-24); 7.84 (d, 8.8, 2H, H-14, H-16); 7.91 (d, 8.8, 2H, H-13, H-17); 7.96 (d, 8.5, 2H, H-8, H-10); 8.06 (d, 8.5, 2H, H-7, H-11); 8.99 (d, 8.0, 1H, H-3).

$^{13}\text{C}$ -NMR (DMSO- $d_6$ ,  $\delta$  ppm): 36.18 (C-18); 54.21 (C-4); 126.32 (C-22); 127.54 (C-8, C-10); 128.13 (C-20, C-24); 128.35 (C-15); 128.65 (C-7, C-11); 128.94

(C-21, C-23); 129.42 (C-13, C-17); 132.88 (C-14, C-16); 137.88 (C-19); 138.63 (C-6); 139.82 (C-9); 142.74 (C-12); 164.95 (C-2); 172.68 (C-5).

RP-HPLC (methanol:water 30:70, v/v; 1 mL/min; 250 nm): purity = 99.63%;  $t_R$  = 4.57 min.

Anal. (%): Calcd. for  $C_{22}H_{18}BrNO_5S$  (488.35 g/mol): C, 54.11; H, 3.72; N, 2.87; S, 6.57. Found: C, 54.06; H, 3.71; N, 2.86; S, 6.57.

Synthesis of 4-benzyl-2-{4-[(4-bromophenyl)sulfonyl]phenyl}-1,3-oxazol-5(4H)-one **4**

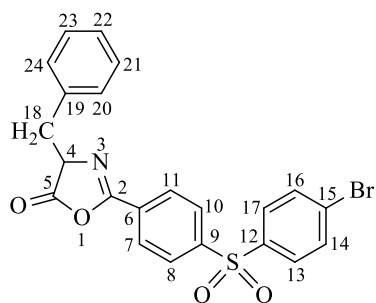


Figure 2.

Atoms numbering of compound **4** used for the assignment of NMR signals

2-{4-[(4-Bromophenyl)sulfonyl]benzamido}-3-phenylpropanoic acid **3** (5.13 g, 10.5 mmol) was suspended with stirring in anhydrous  $CH_2Cl_2$  (50 mL) and an equimolar quantity of 4-methylmorpholine (1.15 mL) was added. Then, 1 mL (10.5 mmol) of ethyl chloroformate was slowly added to the reaction mixture. The solution was stirred for another 30 min and then poured over a mixture of ice and water (100 mL). The organic phase was isolated, washed with 5%  $NaHCO_3$  solution, with water, and dried over anhydrous  $MgSO_4$ . After *in vacuo* concentration, solid product **4** was recrystallized from cyclohexane as white crystals; yield = 93% (4.59 g); m.p. = 161 - 162°C.

UV-Vis ( $CH_3OH$ ,  $\lambda$  nm) (lg  $\epsilon$ ): 202.6 (4.48); 252.9 (4.21).

FT-IR (KBr,  $\nu$   $cm^{-1}$ ): 3086m; 3062m; 3031m; 2988w; 2930w; 2851w; 1822vs; 1651vs; 1600m; 1574s; 1496m; 1473m; 1455m; 1327vs; 1297vs; 1278s; 1159vs; 1044vs; 852s; 614vs; 570vs.

$^1H$ -NMR ( $CDCl_3$ ,  $\delta$  ppm,  $J$  Hz): 3.19 (dd, 14.0, 6.6, 1H, H-18); 3.38 (dd, 14.0, 4.9, 1H, H-18); 4.72 (dd, 6.6, 4.9, 1H, H-4); 7.15-7.24 (m, 5H, H-20 - H-24); 7.67 (d, 8.8, 2H, H-14, H-16); 7.81 (d, 8.8, 2H, H-13, H-17); 7.99 (d, 8.5, 2H, H-8, H-10); 8.04 (d, 8.5, 2H, H-7, H-11).

$^{13}C$ -NMR ( $CDCl_3$ ,  $\delta$  ppm): 37.24 (C-18); 66.84 (C-4); 127.44 (C-22); 128.11 (C-8, C-10); 128.58 (C-7, C-11); 128.92 (C-20, C-24); 129.25 (C-15); 129.50 (C-13, C-17); 129.62 (C-21, C-23); 130.37 (C-6); 132.93 (C-14, C-16); 134.89 (C-19); 139.82 (C-9); 144.90 (C-12); 160.34 (C-2); 176.78 (C-5).

GC-EI-MS ( $m/z$ , rel. abund. %): 425 ( $^{79}Br$ )/427 ( $^{81}Br$ ) (13.36/15.27)  $[M-CO_2]^+$ ; 309/311 (26.72/26.72)  $[^{79}BrC_6H_4SO_2C_6H_4CH_2]^+ / [^{81}BrC_6H_4SO_2C_6H_4CH_2]^+$ ; 203/205 (100, BP/93.13)  $[^{79}BrC_6H_4SO]^+ / [^{81}BrC_6H_4SO]^+$ ; 177 (10.69); 165 (10.69); 155/157 (9.16/6.87)  $[^{79}BrC_6H_4]^+ / [^{81}BrC_6H_4]^+$ ; 152 (3.82); 116 (45.80)  $[C_6H_5C=C=NH]^+$ ; 96 (11.45); 90 (16.79)  $[C_6H_4CH_2]^+$ ; 89 (30.53); 76 (4.20)  $[C_6H_4]^+$ ; 63 (4.58)  $[C_5H_3]^+$ ; 50 (4.58)  $[C_4H_2]^+$ ;  $t_R$  = 56.97 min.

RP-HPLC (methanol:water 60:40, v/v; 1 mL/min; 250 nm): purity = 90.20%;  $t_R$  = 5.52 min.

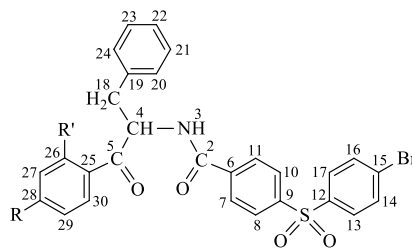
Anal. (%): Calcd. for  $C_{22}H_{16}BrNO_4S$  (470.34 g/mol): C, 56.18; H, 3.43; N, 2.98; S, 6.82. Found: C, 56.25; H, 3.42; N, 2.97; S, 6.84.

Synthesis of 2-{4-[(4-bromophenyl)sulfonyl]benzamido}-3-phenylpropanoyl chloride **5**

2-{4-[(4-Bromophenyl)sulfonyl]benzamido}-3-phenylpropanoic acid **3** (2.69 g, 5.5 mmol) was heated under reflux with an excess amount of thionyl dichloride (10 mL, 16.31 g, 137.09 mmol) until the emissions of gaseous hydrogen chloride and sulphur dioxide ceased. Unreacted  $SOCl_2$  was distilled off under *vacuum* and the obtained yellow crystals was subsequently used without purification; yield = 99% (2.76 g); m.p. = 98 - 100°C;  $C_{22}H_{17}BrClNO_4S$  (506.80 g/mol).

FT-IR (KBr,  $\nu$   $cm^{-1}$ ): 3374m; 3089m; 3064m; 3032m; 2969w; 2948w; 2843w; 1823s; 1794s; 1651s; 1598m; 1573s; 1517m; 1496m; 1472m; 1455m; 1328vs; 1291s; 1161vs; 891m; 850m; 613vs; 574s.

Synthesis of *N*-(1-aryl-1-oxo-3-phenylpropan-2-yl)-4-[(4-bromophenyl)sulfonyl]benzamides **6a,b**



**6a:** R =  $CH_3$ , R' = H; **6b:** R, R' =  $CH_3$

Figure 3.

Atoms numbering of compounds **6a,b** used for the assignment of NMR signals

A 3-fold molar excess of anhydrous aluminium trichloride (2.00 g, 15 mmol) was gradually added under magnetic stirring, at room temperature, to crude 4-benzyl-2-{4-[(4-bromophenyl)sulfonyl]phenyl}-1,3-oxazol-5(4H)-one **4** (2.35 g, 5 mmol) in 25 mL of anhydrous aromatic hydrocarbon (toluene or *m*-xylene). The reaction mixture was further stirred for 20 h until the release of hydrochloric acid ceased and then poured over 100 mL of ice-water mixture acidulated with 5 mL of 37% HCl. The precipitated solid was filtered off, washed with cold water, and then with a cold ethanol-water mixture (1:1, v/v). The

separated aqueous layer was extracted twice with 15 mL of  $\text{CH}_2\text{Cl}_2$ . Combined organic phases were washed with water, dried ( $\text{Na}_2\text{SO}_4$ ), and concentrated by *vacuum* distillation, leaving the second fraction of the raw product. Purification by recrystallization from ethanol gives **6** as colourless crystals.

4-[(4-Bromophenyl)sulfonyl]-N-[1-oxo-3-phenyl-1-(*p*-tolyl)propan-2-yl]benzamide **6a**, obtained by reaction of **4** with toluene (21.63 g, 234.75 mmol); yield = 87% (2.45 g); m.p. = 222 - 224°C.

UV-Vis ( $\text{CH}_3\text{OH}$ ,  $\lambda$  nm) (lg  $\epsilon$ ): 203.5 (4.49); 255.5 (4.26).

FT-IR (KBr,  $\nu$   $\text{cm}^{-1}$ ): 3392s; 3084w; 3060w; 3028m; 2970w; 2930m; 2863w; 1682vs; 1650vs; 1604s; 1573s; 1513vs; 1483s; 1455m; 1329vs; 1299s; 1293s; 1162vs; 855m; 617vs; 578vs.

$^1\text{H-NMR}$  ( $\text{DMSO-d}_6$ ,  $\delta$  ppm,  $J$  Hz): 2.35 (s, 3H,  $\text{CH}_3$ ); 3.01 (dd, 13.9, 8.9, 1H, H-18); 3.19 (dd, 13.9, 4.4, 1H, H-18); 5.66 (m, 1H, H-4); 7.15 (t, 7.4, 1H, H-22); 7.29-7.37 (m, 4H, H-20, H-21, H-23, H-24); 7.32 (d, 8.2, 2H, H-27, H-29); 7.84 (d, 8.5, 2H, H-14, H-16); 7.90 (d, 8.5, 2H, H-13, H-17); 7.93 (d, 8.5, 4H, H-8, H-10, H-26, H-30); 8.04 (d, 8.5, 2H, H-7, H-11); 9.21 (d, 8.0, 1H, H-3).

$^{13}\text{C-NMR}$  ( $\text{DMSO-d}_6$ ,  $\delta$  ppm): 21.25 ( $\text{CH}_3$ ); 36.11 (C-18); 55.84 (C-4); 126.46 (C-22); 127.71 (C-8, C-10); 128.33 (C-15); 128.48 (C-21, C-23, C-27, C-29); 128.80 (C-7, C-11); 129.31 (C-20, C-24); 129.46 (C-13, C-17); 129.58 (C-26, C-30); 132.62 (C-25); 133.04 (C-14, C-16); 137.97 (C-19); 138.60 (C-6); 139.92 (C-9); 142.94 (C-12); 144.02 (C-28); 164.81 (C-2); 197.61 (C-5).

RP-HPLC (methanol:water 60:40, v/v; 1 mL/min; 250 nm): purity = 94.53%;  $t_R$  = 6.82 min.

Anal. (%): Calcd. for  $\text{C}_{29}\text{H}_{24}\text{BrNO}_4\text{S}$  (562.47 g/mol): C, 61.92; H, 4.30; N, 2.49; S, 5.70. Found: C, 61.97; H, 4.29; N, 2.49; S, 5.72.

4-[(4-Bromophenyl)sulfonyl]-N-[1-(2,4-dimethylphenyl)-1-oxo-3-phenylpropan-2-yl]benzamide **6b**, obtained by reaction of **4** with *m*-xylene (21.70 g, 204.39 mmol); yield = 90% (2.59 g); m.p. = 209 - 210°C.

UV-Vis ( $\text{CH}_3\text{OH}$ ,  $\lambda$  nm) (lg  $\epsilon$ ): 203.5 (4.48); 254.6 (4.19).

FT-IR (KBr,  $\nu$   $\text{cm}^{-1}$ ): 3394vs; 3084m; 3063m; 3053m; 3031m; 2962m; 2927m; 2863w; 1682vs; 1654vs; 1610s; 1572vs; 1513vs; 1481vs; 1454s; 1325vs; 1294vs; 1162vs; 835m; 615vs; 578vs.

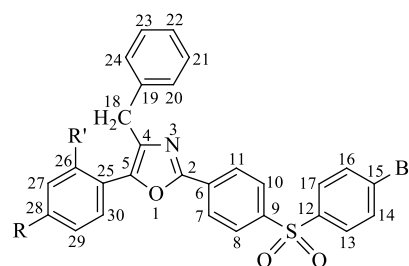
$^1\text{H-NMR}$  ( $\text{DMSO-d}_6$ ,  $\delta$  ppm,  $J$  Hz): 2.27 (s, 3H,  $\text{CH}_3$ ); 2.36 (s, 3H,  $\text{CH}_3$ ); 2.99 (dd, 14.0, 10.0, 1H, H-18); 3.16 (dd, 14.0, 4.4, 1H, H-18); 5.45 (ddd, 10.0, 8.0, 4.4, 1H, H-4); 7.00-7.30 (m, 6H, H-20 - H-24, H-29); 7.09 (bs, 1H, H-27); 7.77 (d, 8.5, 1H, H-30); 7.83 (d, 8.8, 2H, H-14, H-16); 7.90 (d, 8.8, 2H, H-13, H-17); 7.90 (d, 8.5, 2H, H-8, H-10); 8.04 (d, 8.5, 2H, H-7, H-11); 9.15 (d, 8.0, 1H, H-3).

$^{13}\text{C-NMR}$  ( $\text{DMSO-d}_6$ ,  $\delta$  ppm): 20.30 ( $\text{CH}_3$ ); 20.84 ( $\text{CH}_3$ ); 35.36 (C-18); 58.10 (C-4); 126.12 (C-29); 126.26 (C-22); 127.57 (C-8, C-10); 128.12 (C-21, C-23); 128.13 (C-7, C-11); 128.18 (C-30); 128.51 (C-15); 129.02 (C-20, C-24); 129.43 (C-13, C-17); 132.26 (C-27); 132.89 (C-14, C-16); 133.35 (C-25); 137.89 (C-26); 137.97 (C-19); 138.49 (C-6, C-28); 139.77 (C-9); 141.36 (C-12); 164.80 (C-2); 201.33 (C-5).

RP-HPLC (methanol:water 60:40, v/v; 1 mL/min; 250 nm): purity = 92.40%;  $t_R$  = 8.00 min.

Anal. (%): Calcd. for  $\text{C}_{30}\text{H}_{26}\text{BrNO}_4\text{S}$  (576.50 g/mol): C, 62.50; H, 4.55; N, 2.43; S, 5.56. Found: C, 62.56; H, 4.54; N, 2.44; S, 5.54.

Synthesis of 5-aryl-4-benzyl-2-[4-[(4-bromophenyl)sulfonyl]phenyl]-1,3-oxazoles **7a,b**



**7a:** R =  $\text{CH}_3$ , R' = H; **7b:** R, R' =  $\text{CH}_3$

**Figure 4.**

Atoms numbering of compounds **7a,b** used for the assignment of NMR signals

Raw *N*-(1-aryl-1-oxo-3-phenylpropan-2-yl)-4-[(4-bromophenyl)sulfonyl]benzamide **6** (10 mmol) in phosphoryl trichloride (20 mL, 217.83 mmol) was heated at reflux temperature for 4 h. The excess of  $\text{POCl}_3$  was subsequently distilled off *in vacuo*. The oily residue was poured over an ice-water mixture and extracted with  $\text{CH}_2\text{Cl}_2$  (2  $\times$  20 mL). Combined organic phases were washed with 5%  $\text{NaHCO}_3$  solution, then with water and dried over anhydrous  $\text{Na}_2\text{SO}_4$ . The solvent was removed under *vacuum* and the obtained crude solid **7** was recrystallized from ethanol as colourless crystals.

4-Benzyl-2-[4-[(4-bromophenyl)sulfonyl]phenyl]-5-(*p*-tolyl)-1,3-oxazole **7a**, obtained from 5.62 g of **6a**; yield = 87% (4.74 g); m.p. = 211 - 213°C.

UV-Vis ( $\text{CH}_3\text{OH}$ ,  $\lambda$  nm) (lg  $\epsilon$ ): 203.5 (4.49); 252.0 (4.24); 340.1 (4.24).

FT-IR (KBr,  $\nu$   $\text{cm}^{-1}$ ): 3086m; 3065m; 3027m; 2919m; 2860w; 1597s; 1574s; 1542m; 1509s; 1495m; 1473m; 1453m; 1326vs; 1293s; 1158vs; 1089vs; 844s; 618vs; 566vs.

$^1\text{H-NMR}$  ( $\text{CDCl}_3$ ,  $\delta$  ppm,  $J$  Hz): 2.38 (s, 3H,  $\text{CH}_3$ ); 4.18 (s, 2H, H-18); 7.20-7.30 (m, 5H, H-20 - H-24); 7.27 (d, 8.2, 2H, H-27, H-29); 7.54 (d, 8.2, 2H, H-26, H-30); 7.65 (d, 8.6, 2H, H-14, H-16); 7.82 (d, 8.6, 2H, H-13, H-17); 7.99 (d, 8.8, 2H, H-8, H-10); 8.21 (d, 8.8, 2H, H-7, H-11).

$^{13}\text{C}$ -NMR ( $\text{CDCl}_3$ ,  $\delta$  ppm): 21.44 ( $\text{CH}_3$ ); 33.19 (C-18); 125.52 (C-6); 125.96 (C-26, C-30); 126.61 (C-22); 127.10 (C-8, C-10); 128.30 (C-7, C-11); 128.51 (C-27, C-29); 128.73 (C-20, C-24); 128.81 (C-15); 129.34 (C-13, C-17); 129.79 (C-21, C-23); 132.11 (C-28); 132.83 (C-14, C-16); 136.12 (C-25); 138.87 (C-9); 140.62 (C-12); 142.00 (C-4); 142.01 (C-19); 148.20 (C-5); 157.71 (C-2).

RP-HPLC (methanol:water 70:30, v/v; 1 mL/min; 335 nm): purity = 97.70%;  $t_{\text{R}}$  = 4.98 min.

Anal. (%): Calcd. for  $\text{C}_{29}\text{H}_{22}\text{BrNO}_3\text{S}$  (544.46 g/mol): C, 63.97; H, 4.07; N, 2.57; S, 5.89. Found: C, 63.94; H, 4.06; N, 2.58; S, 5.87.

4-Benzyl-2-{4-[(4-bromophenyl)sulfonyl]phenyl}-5-(2,4-dimethylphenyl)-1,3-oxazole **7b**, obtained from 5.77 g of **6b**; yield = 86% (4.81 g); m.p. = 155 - 157°C. UV-Vis ( $\text{CH}_3\text{OH}$ ,  $\lambda$  nm) (lg  $\epsilon$ ): 203.5 (4.49); 249.3 (4.19); 322.5 (4.19).

FT-IR (KBr,  $\nu$   $\text{cm}^{-1}$ ): 3085w; 3065w; 3026m; 2923m; 2857w; 1602m; 1573s; 1558m; 1494m; 1472m; 1454m; 1327vs; 1289m; 1280m; 1158vs; 1101s; 843m; 625s; 570s.

$^1\text{H}$ -NMR ( $\text{CDCl}_3$ ,  $\delta$  ppm,  $J$  Hz): 2.32 (s, 3H,  $\text{CH}_3$ ); 2.38 (s, 3H,  $\text{CH}_3$ ); 3.94 (s, 2H, H-18); 7.07 (bd, 7.6, 1H, H-29); 7.15 (bs, 1H, H-27); 7.20-7.30 (m, 5H, H-20 - H-24); 7.21 (d, 7.6, 1H, H-30); 7.65 (d, 8.7, 2H, H-14, H-16); 7.82 (d, 8.7, 2H, H-13, H-17); 7.99 (d, 8.8, 2H, H-8, H-10); 8.18 (d, 8.8, 2H, H-7, H-11).

$^{13}\text{C}$ -NMR ( $\text{CDCl}_3$ ,  $\delta$  ppm): 20.46 ( $\text{CH}_3$ ); 21.41 ( $\text{CH}_3$ ); 32.46 (C-18); 124.48 (C-6); 126.48 (C-22); 126.78 (C-29); 127.03 (C-8, C-10); 128.32 (C-7, C-11); 128.59 (C-20, C-24); 128.66 (C-21, C-23); 128.80 (C-15); 129.34 (C-13, C-17); 130.17 (C-30); 131.80 (C-27); 132.83 (C-14, C-16); 137.79 (C-26); 137.82 (C-25); 138.96 (C-28); 139.98 (C-9), 140.65 (C-12); 142.01 (C-4, C-19); 148.60 (C-5); 158.43 (C-2).

RP-HPLC (methanol:water 70:30, v/v; 1 mL/min; 335 nm): purity = 99.90%;  $t_{\text{R}}$  = 5.03 min.

Anal. (%): Calcd. for  $\text{C}_{30}\text{H}_{24}\text{BrNO}_3\text{S}$  (558.49 g/mol): C, 64.52; H, 4.33; N, 2.51; S, 5.74. Found: C, 64.57; H, 4.32; N, 2.52; S, 5.76.

#### Cytotoxicity assessment

*Daphnia magna* bioassay was used for the cytotoxicity evaluation. The procedure was described in our previous works [27, 33]. The daphnids were exposed to six concentrations ranging from 3.6 to 130  $\mu\text{g}/\text{mL}$ , for a period of 48 h. The test was carried out in duplicate, and phenylalanine, compound **1**, and 1% DMSO were used as controls. The bioassay was performed in a controlled environment at  $25 \pm 1^\circ\text{C}$ , 75% RH (climatic chamber Sanyo MLR-351 H, USA).

At 24 and 48 h, the lethality was recorded, and the LC50 values were calculated by interpolation on the lethality curves. GUSAR software was applied for the prediction of LC50<sub>48h</sub>.

## Results and Discussion

### Chemistry

Synthetic transformations into title derivatives started from the following raw materials: 4-[(4-bromophenyl)sulfonyl]benzoic acid **1** and corresponding acyl chloride **2** which are literature-known compounds [31, 32]. Carboxylic acid **1** was obtained from commercially available 4-methylbenzene-1-sulfonyl chloride and bromobenzene by two successive reactions [31]. Then, compound **1** was converted into 4-[(4-bromophenyl)sulfonyl]benzoyl chloride **2**, the method being adapted based on previously described procedures [12, 30]. Further, phenylalanine was acylated with raw material **2** to new 2-{4-[(4-bromophenyl)sulfonyl]benzamido}-3-phenylpropanoic acid **3**. Subsequently, *N*-acyl- $\alpha$ -amino acid **3** was cyclodehydrated, under the action of ethyl carbonochloridate in presence of 4-methylmorpholine to 4-benzyl-2-{4-[(4-bromophenyl)sulfonyl]phenyl}-1,3-oxazol-5(4*H*)-one **4**. The 2-{4-[(4-bromophenyl)sulfonyl]benzamido}-3-phenylpropanoyl chloride **5** was prepared by reacting *N*-acylated phenylalanine **3** with thionyl dichloride. By  $\text{AlCl}_3$ -catalyzed acylation of arenes (toluene, *m*-xylene) with 2-aryl-4-benzyl-1,3-oxazol-5(4*H*)-one **4**, the 4-aryl-2-aza-3-benzyl-1-[(4-bromophenyl)sulfonyl]phenyl]-1,4-butanediones **6a,b** were obtained. Further, the intramolecular condensation and dehydration of *N*-(1-aryl-1-oxo-3-phenylpropan-2-yl)-4-[(4-bromophenyl)sulfonyl]benzamides **6a,b** under the action of phosphoryl trichloride, known as Robinson-Gabriel reaction, yielded 5-aryl-4-benzyl-1,3-oxazoles containing 4-[(4-bromophenyl)sulfonyl]phenyl moiety in 2-position **7a,b**.

Depicted chemical structures of new derivatives were established based on NMR spectroscopy ( $^1\text{H}$ ,  $^{13}\text{C}$ , COSY, HETCOR), UV-Vis, IR, MS data and elemental analysis.

The UV-Vis spectra of the synthesized compounds presented E band at  $\lambda_{\text{max}} = 202.6$  or 203.5 nm, and B band in 249.3 - 255.5 nm range. In addition, 1,3-oxazoles **7** spectra showed a third absorption maximum at longer wavelengths (322.5 or 340.1 nm), due to the appearance of the heterocyclic chromophore, which determines the extent of  $\pi$  electrons conjugation.

The FT-IR spectra of the new compounds presented all the expected characteristic absorption bands. In the case of acyclic precursors **3** and **6a,b**, main peaks were due to N-H valence vibration,  $\nu(\text{N-H})$ , in 3356 - 3394  $\text{cm}^{-1}$  range, ketonic carbonyl valence vibration,  $\nu(\text{O}=\text{C}-\text{C})$ , at 1682 or 1734  $\text{cm}^{-1}$ , and stretching vibration of amidic carbonyl,  $\nu(\text{O}=\text{C}-\text{N})$ , in 1622 - 1654  $\text{cm}^{-1}$  interval. Representative peaks for the hydrogen-bonded dimer of *N*-acyl phenylalanine **3** were also recorded: a broad absorption band due to O-H stretch,  $\nu(\text{O-H})$ , between 2500 and 3300  $\text{cm}^{-1}$ , and four broad bands in 2530 - 2703  $\text{cm}^{-1}$  interval. The IR spectrum of acyl chloride **5** showed a fundamental band due

to  $\nu(\text{O}=\text{C}-\text{Cl})$  at  $1823\text{ cm}^{-1}$ , Fermi resonance peak at  $1794\text{ cm}^{-1}$ , and band due to C-Cl valence vibration,  $\nu(\text{Cl}-\text{C}=\text{O})$ , at  $891\text{ cm}^{-1}$ . The IR spectral data of five-membered-ring systems **4** and **7a,b** demonstrated that cyclocondensation of open-chain intermediates **3** and **6a,b**, respectively occurred. In compound **4** IR spectrum, carbonyl valence vibration,  $\nu(\text{C}=\text{O})$ , was shifted at higher wavenumber ( $1822\text{ cm}^{-1}$ ); besides,  $\nu(\text{N}-\text{H})$ ,  $\nu(\text{O}-\text{H})$ , and  $\nu(\text{O}=\text{C}-\text{N})$  peaks were not recorded. Also, absorption bands were not observed in the N-H and C=O regions of 1,3-oxazoles **7** spectra. The IR spectra of pentaatomic heterocycles **4** and **7a,b** showed C=N valence vibration,  $\nu(\text{C}=\text{N})$ , at  $1651\text{ cm}^{-1}$  (**4**), and  $1597$  or  $1602\text{ cm}^{-1}$  (**7a,b**). Furthermore, C-O-C symmetrical stretching vibration,  $\nu_{\text{sym}}(\text{C}-\text{O}-\text{C})$ , appeared at  $1044\text{ cm}^{-1}$  in saturated azlactone **4** spectrum, and at  $1089\text{ cm}^{-1}$  or  $1101\text{ cm}^{-1}$  in 1,3-oxazoles **7a,b** spectra, and C-O-C asymmetrical stretching vibration,  $\nu_{\text{as}}(\text{C}-\text{O}-\text{C})$ , was registered at  $1278\text{ cm}^{-1}$  (**4**), and  $1280\text{ cm}^{-1}$  (**7b**).

The  $^1\text{H-NMR}$  spectra of the new compounds confirmed proposed chemical structures. Moreover,  $^1\text{H}-^1\text{H}$  COSY spectra facilitated assigning the signals. In  $^1\text{H-NMR}$  spectra of compounds **3** and **6a,b**, H-3 signal was recorded as a doublet at  $\delta$  values of  $8.99\text{ ppm}$  (**3**) and  $9.15$  or  $9.21\text{ ppm}$  (**6a,b**), due to the coupling with H-4 ( $^3J = 8.0\text{ Hz}$ ). As proof that cyclization of acyclic intermediates **3** and **6a,b** have taken place, in  $^1\text{H-NMR}$  spectra of heterocycles **4** and **7a,b**, the signal assigned to NH proton (observed in the structure of the two raw materials) was absent. In  $^1\text{H-NMR}$  spectra of **3** and **6a,b**, the H-4 signal appeared as a doublet of doublets or multiplet at  $\delta = 4.64\text{ ppm}$  (**3**) and  $5.45$  or  $5.66\text{ ppm}$  (**6a,b**), due to H-4 coupling with the two nonequivalent H-18 protons and one NH proton. For azlactone **4**, the H-4 signal was registered at  $\delta = 4.72\text{ ppm}$  as a doublet of doublets due to coupling with the nonequivalent H-18 protons. The signal of H-4 (that appeared in  $^1\text{H-NMR}$  spectra of **6a,b**), was not observed in spectra of heterocyclic analogues **7a,b** as proof that cyclocondensation occurred. The  $^1\text{H-NMR}$  spectrum of **4** displayed two signals due to nonequivalent H-18 protons as a doublet of doublets, which showed discernible downfield shifts of  $0.15$  and  $0.18\text{ ppm}$ , compared to corresponding protons signals in the spectrum of intermediate **3**. The  $^1\text{H-NMR}$  spectra of 1,3-oxazoles **7a,b** highlighted a downfield shift of signal assigned to two magnetically equivalent H-18 protons as a singlet at  $\delta = 3.94$  or  $4.18\text{ ppm}$  relative to the two signals registered for *N*-acyl- $\alpha$ -amino ketones **6a,b** as a doublet of doublets (due to germinal coupling between methylene nonequivalent protons and vicinal coupling between H-18 protons and H-4 proton) at  $3.01$  and  $3.19\text{ ppm}$  (**6a**) and  $2.99$  and  $3.16\text{ ppm}$  (**6b**).

The  $^{13}\text{C-NMR}$  spectra also proved the syntheses of the compounds took place. Besides,  $^1\text{H}-^{13}\text{C}$  COSY spectra allowed the assignments of  $^{13}\text{C}$  peaks. The C-4 signal,

which was recorded at  $\delta = 54.21\text{ ppm}$  in the  $^{13}\text{C-NMR}$  spectrum of *N*-acylated phenylalanine **3**, was shifted downfield with  $12.63\text{ ppm}$  after cyclization to **4**.

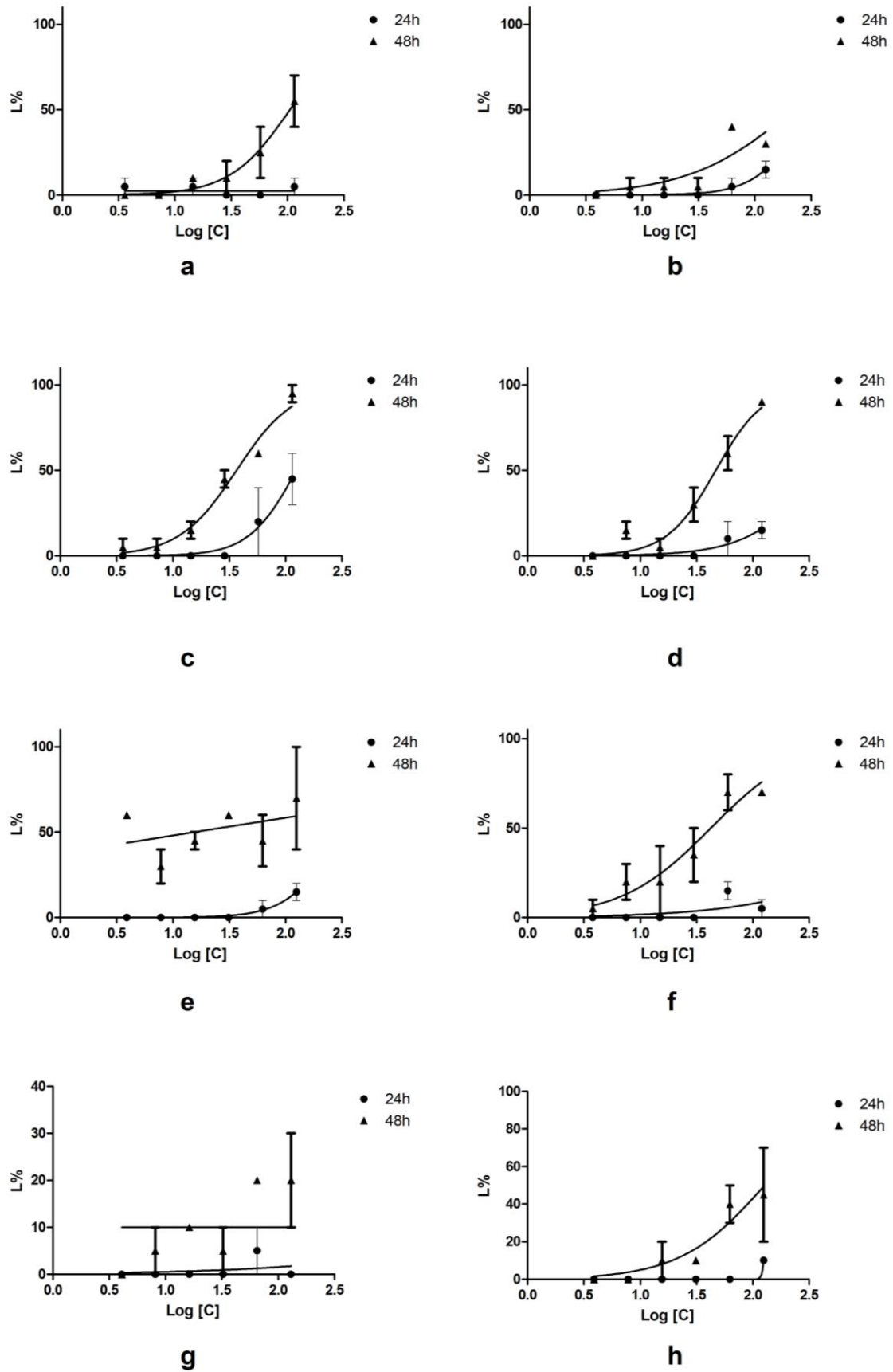
Moreover, in the case of 1,3-oxazol-5(4*H*)-one **4**, the C-2 resonated at  $160.34\text{ ppm}$  (being shifted upfield with  $4.61\text{ ppm}$  than C-2 of intermediate **3**), and the C-5 at  $176.78\text{ ppm}$  (being deshielded with  $4.10\text{ ppm}$  than C-5 of precursor **3**). The signal assigned to C-2 of 2,4,5-trisubstituted 1,3-oxazoles **7a,b** was registered at  $\delta = 157.71$  or  $158.43\text{ ppm}$ , while the peak attributed to C-2 from -CONH- group of *N*-acyl- $\alpha$ -amino ketones **6a,b** (from  $\delta = 164.80$  or  $164.81\text{ ppm}$ ) was absent in aromatic heterocycles **7a,b** spectra. C-5 of 1,3-oxazoles **7a,b** resonated at  $\delta = 148.20$  or  $148.60\text{ ppm}$ , whereas the corresponding atom of **6a,b** at  $\delta = 197.61$  or  $201.33\text{ ppm}$ , revealing a high-field shift for this carbon from cyclization products **7a,b** structure, as a confirmation that the reaction took place.

The mass spectrum of 1,3-oxazol-5(4*H*)-one **4** obtained by GC-EI-MS analysis played an important role in elucidating the structure of this compound. The two very unstable molecular ions of **4** that correspond to bromine isotopes ( $^{79}\text{Br}/^{81}\text{Br}$ ) did not show peaks in the mass spectrum. They began to fragment at the level of the heterocyclic nucleus by removing a molecule of carbon dioxide with the obtaining of two cation-radicals corresponding to isotopes of bromine with  $m/z = 425/427$  (with relative abundances:  $13.36\%$  and  $15.27\%$ , respectively).

The base peak (PB)  $[^{79}\text{BrC}_6\text{H}_4\text{SO}]^+$  with  $m/z = 203$  and corresponding cation  $[^{81}\text{BrC}_6\text{H}_4\text{SO}]^+$  with  $m/z = 205$  (with relative abundance =  $93.13\%$ ) were formed in accordance with  $^{79}\text{Br}/^{81}\text{Br}$  isotopic ratio of about 1:1. Other main fragments of **4** corresponding to  $^{79}\text{Br}$  and  $^{81}\text{Br}$  isotopes are indicated in section Materials and Methods.

#### Cytotoxicity assessment

*Daphnia magna* bioassay results are presented in Table I. At 24 h, at the highest concentrations, with the exception of **6a** which exhibited at 24 h an L% of  $45\%$ , all tested compounds did not exceed  $20\%$ . At 48 h, the highest toxicity was induced by compound **7a**. However, this toxicity wasn't dependent on the concentrations. For this compound, at all concentrations was observed an L% between  $30$  and  $60\%$ , and thus, lower goodness of fit. For all other compounds, the correlation between the concentrations and the L% was satisfactory (for compound **4** and phenylalanine between  $0.6$  and  $0.69$  and for other compounds over  $0.7$ ) (Figure 5). The cytotoxicity on *D. magna* decrease in the following order: **7a**, **6a**, **7b**, **6b**, **3**, **1**, and **4**. Phenylalanine induced a maximum of  $30\%$  lethality at 48 h. The 95% CI values for compounds **6a**, **7b** and **6b** indicate similar toxicities of these three compounds. The values obtained experimentally were significantly higher than those predicted using GUSAR software.



**Figure 5.**  
Lethality curves on *Daphnia magna* for selected compounds;  
a – 3, b – 4, c – 6a, d – 6b, e – 7a, f – 7b, g – phenylalanine, h – 1

**Table I**  
Results of *Daphnia magna* bioassay

Compounds	Predicted LC50 <sub>48 h</sub> (µg/mL)	Determined LC50 <sub>48 h</sub> (µg/mL)	95% CI of LC50 <sub>48 h</sub> (µg/mL)
<b>3</b>	0.44	105.00	72.59 - 151.80
<b>4</b>	0.28	222.2	83.99 - 587.80
<b>6a</b>	0.11	36.67	30.37 - 44.27
<b>6b</b>	0.04	46.85	37.80 - 58.07
<b>7a</b>	0.03	~ 15.49	NC
<b>7b</b>	0.02	42.34	26.08 - 68.74
phenylalanine	170.79	NC	NC
<b>1</b>	11.80	127.3	63.08 - 256.9

LC50 – 50% lethal concentration; 95% CI – 95% confidence interval; NC – not calculated due to obtained results

## Conclusions

Seven new compounds, derived from phenylalanine, which incorporate in the structure the 4-[(4-bromophenyl)sulfonyl]phenyl fragment, were designed, synthesized, and physicochemically characterized. 1,3-Oxazol-5(4*H*)-one **4** was produced by Steiger acylation of phenylalanine with acyl chloride **2**, followed by intramolecular cyclodehydration of acyclic precursor **3**. *N*-Acyl phenylalanyl chloride **5** was obtained by reaction of **3** with SOCl<sub>2</sub>. *N*-Acyl- $\alpha$ -amino ketones **6a,b** were generated by the reaction of 1,3-oxazol-5(4*H*)-one **4** with aromatic hydrocarbons in presence of aluminium trichloride. The 2,4,5-trisubstituted 1,3-oxazoles **7a,b** were synthesized by cyclization of open-chain intermediates **6a,b** under the action of phosphoryl trichloride. Compounds structure was elucidated through spectral methods and elemental analysis.

*Daphnia magna* bioassay revealed that the newly synthesized compounds **6a**, **6b**, **7a**, and **7b** exhibited high cytotoxicity being promising candidates for future biological investigations.

## Conflict of interest

The authors declare no conflict of interest.

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