

EVALUATING THE CAPACITY OF HUMAN SERUM ALBUMIN TO REDUCE NON-SPECIFIC BINDING OF MELOXICAM IN THE ULTRAFILTRATION PROCESS

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Abstract

Due to the importance that plasma protein binding of drugs plays and the constant interest to optimize study methods of the process, this study aimed to test the capacity of a human serum albumin solution to reduce non-specific binding of highly bound meloxicam, based on the ability of the protein to block most non-specific binding sites of the ultrafiltration devices. Samples of meloxicam and internal standard piroxicam were prepared in both phosphate-buffered saline solution, commonly used in biological research, and 5% human serum albumin solution. After the ultrafiltration process, the free drug fraction was analysed using a reversed phase HPLC-UV method. A decrease in the concentration of meloxicam after ultrafiltration in both matrices was observed, the decrease being higher in the case of phosphate-buffered saline solution. The mean determined degree of binding to albumin for meloxicam was 55%. The high decrease in concentration for meloxicam in the non-proteic matrix indicates a very high degree of non-specific binding, phenomenon which appears to be indeed reduced for the samples prepared in human serum albumin solution. The study also emphasises the very close attention that should be paid to the preparation and processing of samples in protein matrices and to the experimental conditions.

Rezumat

Datorită importanței pe care legarea medicamentelor de proteinele plasmatice o prezintă și a interesului constant de a optimiza metodele de studiu ale procesului, acest studiu a avut ca scop testarea capacității unei soluții de albumină serică umană de a reduce legarea nespecifică a meloxicamului puternic legat, pe baza capacității proteinei de a bloca majoritatea situsurilor de legare nespecifică a dispozitivelor de ultrafiltrare. Probele de meloxicam și standard intern piroxicam au fost preparate atât în soluție de tampon fosfat salin, utilizată în mod obișnuit în cercetări biologice, cât și în soluție de albumină serică umană 5%. După procesul de ultrafiltrare, fracția de medicament liber a fost analizată folosind o metodă HPLC-UV în fază inversă. S-a observat o scădere a concentrației de meloxicam după ultrafiltrare în ambele matrici, scăderea fiind mai mare în cazul soluției de tampon fosfat salin. Gradul mediu de legare de albumină determinat pentru meloxicam a fost de 55%. Scăderea ridicată a concentrației de meloxicam în matricea non-proteică indică un grad foarte ridicat de legare nespecifică, fenomen care pare a fi într-adevăr redus pentru probele preparate în soluție de albumină serică umană. Studiul scoate în evidență, de asemenea, atenția mare care trebuie acordată condițiilor experimentale, preparării și prelucrării probelor în matrici proteice.

Keywords: ultrafiltration, non-specific binding, albumin, meloxicam, protein binding

Introduction

Plasma protein binding (PPB) has a critical impact on the pharmacokinetic and pharmacodynamic properties of substances, influencing also their toxicity, which may increase as a result of displacement interactions [2, 3, 7, 12, 22, 33, 37].

In general, PPB is a reversible phenomenon, being governed by the law of mass action, which ultimately leads to a balance between the bound and free drug fractions. It is commonly accepted that in the absence of a specific transport system, only free drug molecules

are able to diffuse through membranes and can be distributed to the tissues leading to a pharmacological effect [2, 7, 12, 23, 37]. Regarding the binding of exogenous compounds, the most important plasma proteins involved in the process are human serum albumin (HSA) and alpha-1-acid glycoprotein (AGP), followed by lipoproteins and globulins. HSA has an extraordinary capacity to reversibly bind ligands that are mainly acidic, but also neutral [9, 15]. HSA has multiple binding sites, but, among these, only two are specific for exogenous substances. These are known as the Sudlow I and Sudlow II site or the warfarin site

(site I) and the benzodiazepine site (site II), depending on the substances for which they are specific [2, 9, 12, 15, 36]. A third site for which the existence of a specificity for digitoxin binding was demonstrated, is mentioned in the literature, but it has been studied in a lower manner [6, 28, 39]. AGP is an acute phase protein with a lower binding capacity than albumin, but a higher affinity for ligands, which makes the dissociation of the bound drug molecules more difficult. Although it has a very high affinity for basic substances, some neutral or acidic compounds and endogenous steroid hormones such as progesterone, can also bind to it. Multiple binding sites have been reported in the case of AGP, but only one of them has been shown to be important for the process of drug binding [10, 12, 15].

Regarding the study of PPB, various methods were developed and described in the literature, but from the perspective of drug development, only equilibrium dialysis and ultrafiltration (UF) are widely accepted [3, 5, 12, 15, 20, 33, 34, 37, 38].

UF uses a semipermeable membrane for the physical separation of bound and free drug fractions, and it is considered to be one of the fastest and simplest methods for determining the free fraction of a drug which could also be used during clinical monitoring [3, 12, 20, 30, 38].

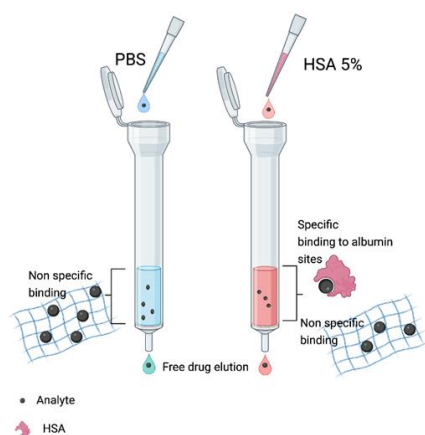


Figure 1.

Experimental *in vitro* protocol in order to assess NSB (created with BioRender.com)

The highest disadvantage of the UF method, however, is the non-specific binding (NSB) of substances to the semi-permeable membrane and to the compartment of the UF device, which provide NSB sites due to their characteristic polarity and charge [3, 12, 16, 20, 30, 35, 38]. There is a constant interest regarding the optimization of the UF method in order to overcome this limitation [1, 8, 11, 13, 16, 17, 29, 30, 35]. In a study by Wang C and Williams NS [35], which had a mass balance approach, it was observed that the behaviour of some compounds in terms of NSB is very different when they are incubated with the UF

device in plasma compared to phosphate buffered saline (PBS), commonly used in biological research (Figure 1). This was thought to be a consequence of the fact that serum proteins can adsorb on the surface of different types of materials, leading to the blockage of most NSB sites. The study showed that, using this approach, the limitation given by NSB can be overcome even in the case of compounds with increased lipophilia.

In the present study, we aimed to assess whether a human serum albumin solution has the capacity to reduce NSB using meloxicam (MXC) and piroxicam (PXC) as reference substances. MXC and PXC are both non-steroidal anti-inflammatory drugs [25, 26] that have a high affinity for plasma proteins, especially albumin [4, 18, 19, 27, 31]. The interaction between meloxicam and HSA is strong in nature, occurring at a single site with high affinity, which is found in the IIA sub-domain and has an association constant of the order of 10^5 M^{-1} [27, 31, 32]. Meloxicam is bound to human serum albumin mainly by hydrophobic interactions, but hydrogen bonds cannot be excluded either. In the case of piroxicam, studies have shown that it binds to albumin in the IIA subdomain, where tryptophan and tyrosine residues are present, but in addition to this, it may also bind to tyrosine residues located in the IB, IIB or IIIA subdomains [19, 24]. Binding of piroxicam to human serum albumin is a spontaneous and exothermic process involving the presence of hydrogen bonds and hydrophobic interactions. The association constant is also of the order of 10^5 M^{-1} [4, 21]. The choice of MXC - PXC combination was made based on the fact that often the two substances are used in tandem analyte - internal standard for quantification from biological samples using chromatographic techniques [14]. In the present study, piroxicam was used as an internal standard (ISTD).

Materials and Methods

Reference substances, reagents and solvents

The substances were obtained as follows: MXC (working standard) was obtained from a local pharmaceutical company, while PXC (working standard) was obtained from Nantong Jinghua Pharmaceuticals Co LTD (China). Acetonitrile, methanol, phosphoric acid 85%, sodium chloride, potassium chloride, disodium phosphate, potassium dihydrogen phosphate were purchased from Merck (Germany), hydrochloric acid 32% was purchased from Micromchim (Romania) and sodium hydroxide from Chemapol (Czech Republic), all of the reagents had appropriate analytical purity. Human Albumin Grifols 200 g/L, solution for infusion, was purchased from Instituto Grifols SA (Spain).

Chromatographic conditions

The conditions for the chromatographic analysis were established after a slight modification of a previously

published method by Imre S *et al.* [14]: the same HPLC system Agilent Technologies 1100 Series type was used and the analytical column was Zorbax SB-C18 (Solvent Saver Plus 100 x 3 mm, 3.5 μ m). The column temperature was set at 45°C. The mobile phase contained a mixture of acetonitrile (ACN) and 15 mM phosphoric acid (H₃PO₄) solution, according to the following gradient: 0 - 1 min isocratic elution 77% H₃PO₄ 15 mM, 23% ACN; 1 - 6 min gradient elution 77% \rightarrow 60% H₃PO₄ 15 mM, 23% \rightarrow 40% ACN; 6 - 7 min restoring the initial composition 60% \rightarrow 77% H₃PO₄ 15 mM, 40% \rightarrow 23% ACN; 7 - 10 min re-equilibration step 77% H₃PO₄ 15 mM, 23% ACN. The flow rate was 1.7 mL/min, sample temperature 20°C and the injection volume 40 μ L. The detection wavelength was set at 364 nm.

Preparation of stock and standard solutions

Stock solution of MXC and PXC. The final stock solutions of MXC and PXC with a concentration of 6 μ g/mL were prepared by the corresponding dilution with PBS (pH 7.40) of an initial solution, with a concentration of 300 μ g/mL prepared in methanol. The concentration of methanol in the final stock solutions was 2% (v/v).

Standard solutions. Seven calibration standards with increasing concentrations of MXC (15, 30, 150, 900, 1500, 2100, 2400 ng/mL) and a constant concentration of PXC as an ISTD (600 ng/mL) were prepared by dilution of the final stock solutions with PBS (pH 7.40) and 5% HSA solution, respectively. The 5% HSA solution was prepared by the corresponding dilution of the 20% infusion solution with purified water.

Standard solutions for testing method accuracy and precision and for the UF study. Three levels of standard solutions were prepared in PBS (pH 7.40) and 5% HSA solution with the concentrations for MXC of 900, 1500 and 2400 ng/mL, respectively, and PXC as an ISTD at a concentration of 600 ng/mL.

Experimental UF protocol

Amicon Ultra-2 centrifugal filter units (Ultracel-10K regenerated cellulose membrane, 2 mL) were purchased from Merck (Germany). The ultrafiltration tubes were weighted before and after the UF process. UF of 1000 μ L samples was performed at room temperature, applying a relative centrifugal force (RCF) of 5000 g, for 10 minutes. Two series of MXC and PXC solutions prepared in PBS and 5% HSA solution, respectively, were subjected to parallel UF. The three concentration levels chosen for MXC correspond in order of magnitude to the plasma levels achieved after administration of a single dose of 20 mg MXC: 900, 1500 and 2400 ng/mL. The concentration of PXC in the analysed samples was 600 ng/mL.

Processing of samples and standard solutions for HPLC analysis

For the HPLC analysis, the samples and standard solutions prepared in 5% HSA were subjected to

precipitation with a solution of 20% perchloric acid (HClO₄), the ratio of 5% HSA solution:HClO₄ 20% being 20:3. After the addition of HClO₄ 20%, the solutions were vortexed for one minute, left at rest for 5 minutes and centrifuged for 10 minutes at 10000 rpm, after which the supernatant was subjected to analysis. The same treatment was applied to the samples in 5% HSA solution both before and after UF, but also to the samples and standard solutions prepared in PBS.

Results and Discussion

Quality parameters of the applied analytical method

The experimental analytical approach was based on the modification of an HPLC method with UV detection previously developed and validated by Imre S *et al.* [14]. A column with the same geometric characteristics and particle size was used, but with C18 stationary phase, instead of C8, and a 15 mM phosphoric acid solution for the mobile phase, instead of phosphate buffer. Taking into account the fact that the difference in stationary phase is only in terms of hydrophobicity and the pH difference of the aqueous mobile phase is in a domain in which MXC and PXC are in the same ionization form, as acid ($pK_{aMXC} = 4.2$; $pK_{aPXC} = 5.5$), and the same equipment was used, it was considered that the modified method can be applied in the present study verifying its analytical performance in terms of specificity, linearity, accuracy and precision. All these parameters proved to be appropriate as it follows.

Specificity. MXC and PXC (ISTD) are specifically separated at retention times of 5.199 (\pm 0.073) min and 2.755 (\pm 0.105) min, respectively, in both PBS (pH 7.4) and 5% HSA solution.

Linearity. The calibration equation in PBS was Area ratio = 0.0018 c + 0.0141 (N = 7) and the correlation coefficient was $r > 0.999$. The residuals showed a random variation, being below 4% in absolute values. In the case of standard solutions in 5% HSA, the calibration equation was Area ratio = 0.00111 c - 0.15442 with a correlation coefficient $r > 0.998$ over 900 - 2400 ng/mL concentration range, a relevant biological concentration domain. The residuals, as well, showed a random variation, being below 4% in absolute values.

Accuracy and precision. The relative standard deviation DSR% and the relative error Er% for intra-series (n = 3) precision and accuracy of the method had absolute values below 2% for determinations made in PBS, suitable for analytical determinations, and below 10%, for those in 5% HSA, adequate for bio-analytical determinations.

Relative recovery of the analyte. The mean relative recovery of MXC was 98.38% (\pm 4.30).

Qualitative study of the non-specific binding of MXC and PXC to the ultrafiltration device used

Taking into account data from the literature on the NSB of substances to the ultrafiltration device materials,

as well as indicated methods that could reduce this phenomenon, it was studied whether a human serum albumin solution would resolve possible NSB of meloxicam and piroxicam compared to PBS. As suggested and demonstrated by Wang C and Williams NS [35] serum proteins can adsorb on the surface of different types of materials, leading to the blockage

of most NSB sites. The device used in the present study was made of styrene-butadiene copolymer with a filter membrane of regenerated cellulose.

Figures 2 and 3 show comparative chromatograms of a 2400 ng/mL MXC sample before and after UF in PBS and 5% HSA solution, respectively.

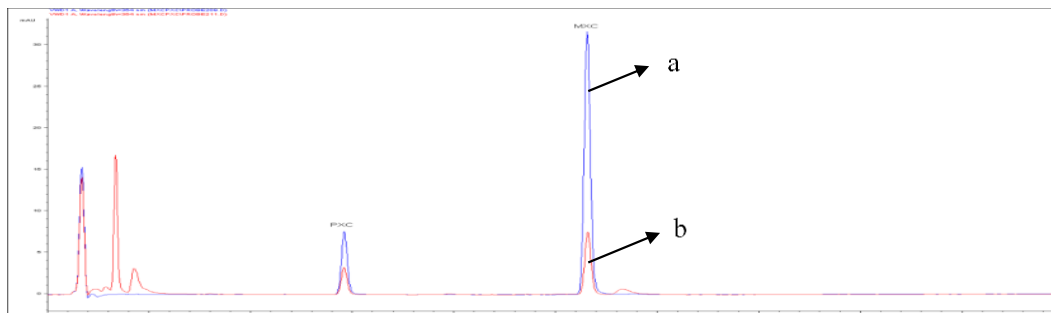


Figure 2.

Chromatograms of a 2400 ng/mL MXC sample (PXC 600 ng/mL) in: a) PBS, b) 5% HSA, before UF

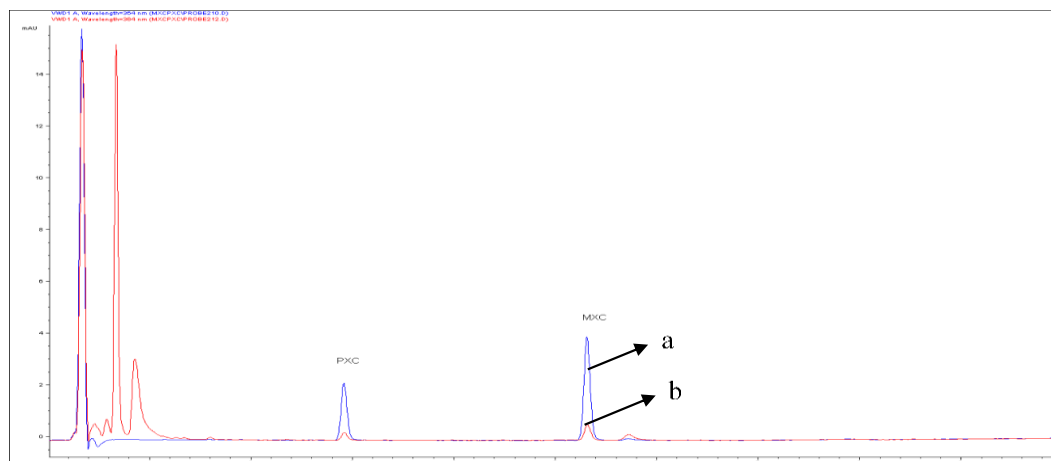


Figure 3.

Chromatograms of a 2400 ng/mL MXC sample (PXC 600 ng/mL) in: a) PBS, b) 5% HSA, after UF

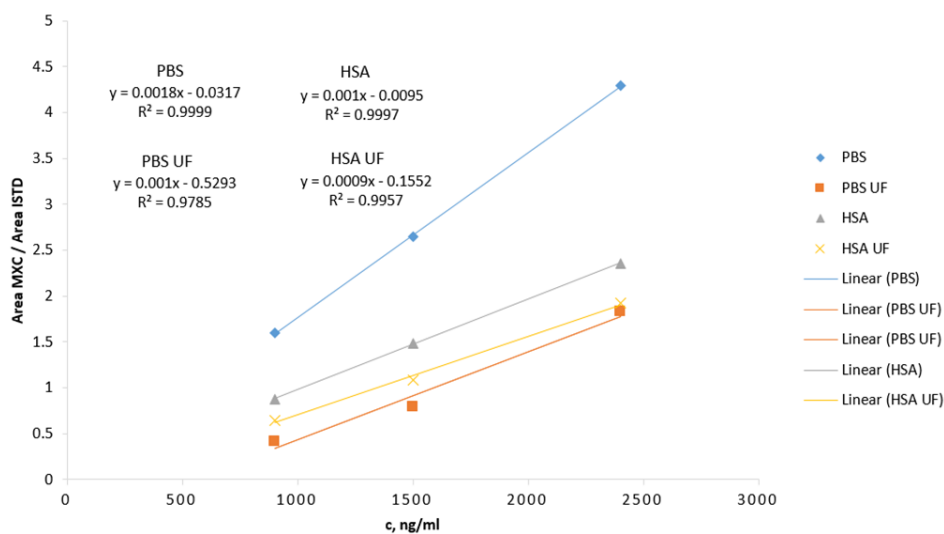


Figure 4.

Correlation of MXC/PXC area ratio with theoretical concentration level for MXC samples prepared in PBS and 5% HSA, before and after UF

Table I

Assay of MXC and PXC samples in the two matrices considered, before and after UF

Before/After UF	Matrix	MXC		PXC		Area ratio MXC/PXC		
		C (ng/mL)	Area	C (ng/mL)	Area			
Before UF	PBS	900	51.1	600	32	1.597		
After UF			2.3		5.6	0.411		
Before UF	HSA		11.4		13.1	0.870		
After UF			0.84		1.3	0.646		
Before UF	PBS		1500		85.5	600	32.3	2.647
After UF					5.3		6.7	0.791
Before UF	HSA	19.9		13.4	1.485			
After UF		1.3		1.2	1.083			
Before UF	PBS	2400		137.7	600		32.1	4.290
After UF				17.2			9.4	1.830
Before UF	HSA		32.7	13.9		2.353		
After UF			2.5	1.3		1.923		

The aim of the study was to provide an estimation of the behavior of MXC and PXC regarding both non-specific and HSA binding, thus singlicate measurements were used.

From the data collected after the experimental assay, several qualitative interpretations can be made: there is a decrease in concentration for MXC and PXC after UF compared to the initial solution, both in PBS and HSA solution (Figures 2 and 3); the decrease in concentration is much more pronounced in the case of PBS solutions compared to HSA, which indicates that indeed HSA blocks NSB sites to some extent. Even in terms of area ratio, there is a decrease after UF, which indicates a different behaviour of the two substances both in terms of NSB, but also in terms of binding to albumin (Table I). The assignment of a linear mathematical model to the experimental points shown in Figure 4 indicates a very good

correlation in the case of solutions prepared in PBS and HSA before UF but it is observed that after UF, the linearity is somewhat reduced for the solutions prepared in HSA and is no longer appropriate in the case of PBS (correlation coefficient < 0.99). Therefore, the PBS solution, at least in the case of the substances considered, appears to be an unfavourable factor in terms of NSB compared to albumin. Albumin acts favourably from this point of view, probably blocking at least some of the binding sites, thus diminishing the NSB.

Determination of the degree of binding of MXC to human serum albumin under the tested experimental conditions

Under the experimental conditions of the study, we assessed a reproducible degree of binding of MXC to HSA of about 55% (Table II), determined in singlicate at three levels of concentration.

Table II

Recovery of MXC after UF in 5% HSA solution

Before/After UF	C (ng/mL)	m _{sol} , g	m _{MXC} , ng	Recovery %	Mean Recovery % (DSR%)	Degree of binding
Before UF	923.1	0.9922	915.9	44.13	45.16 (6.98)	54.84
After UF	721.2	0.5605	404.2			
Before UF	1477	0.9958	1470.8	42.64		
After UF	1115.1	0.5625	627.2			
Before UF	2258.5	0.9883	2232	48.69		
After UF	1871.6	0.5807	1086.8			

The determined binding degree of MXC to HSA is well below the value of over 99% reported in the literature [27, 31]. Possible explanations for this result: it was used standardized HSA solution, a much simplified matrix compared to real human plasma; the presence of other compounds in the HSA infusion solution used as a dissolution medium (sodium caprylate, sodium N-acetyltryptophanate) could have influence on the binding of the drugs to HSA; the experiment was performed entirely at room temperature; the relative centrifugal force was increased during the preliminary tests to 5000 g from 1000 g, used in other studies in the literature, because filtration was not possible at lower speeds for the considered device,

which could have affected the binding equilibrium. However, the DSR% of 6.98 reveals a reproducible protein binding over a relatively wide concentration domain.

The results of the present study are related to the following issues. There are many physicochemical methods used in laboratory practice that limit NSB (change in pH, electrolyte concentration, co-use of additional compounds – proteins, non-ionic surfactants) [30]. These methods can only be limitedly used in biological matrices without affecting the real characteristics of living systems (for example, biological constants of plasma: pH = 7.4, osmotic colloid pressure, type and concentration of electrolytes). On the other hand,

the use of *in vitro* experimental models can provide extremely useful information, but these models are based on a rather crude approximation of biological reality (as in this case, the use of a 5% HSA solution). Useful experiments can be performed using *ex vivo* models, in which whole blood is processed by centrifugation containing the previously administered analyte (by oral or parenteral route). However, the treatment of the sample can add important differences: the plasma obtained contains many proteins, including 55 to 60% albumin, but also globulins around 40%, as well as lipoproteins, enzymes, coagulation factors etc. If serum (obtained by centrifugation after blood coagulation) is used, the composition of the biological matrix will be different from plasma in the absence of fibrinogen and most of the coagulation factors. In addition, in the case of the use of whole blood, the binding of some analytes to the blood cells (such as erythrocyte membrane) may occur.

The more complicated the biological model used is, the closer is to the biological reality of the living organism, but the variables are more difficult to control, especially in the case of equilibrium systems, such as binding to plasma proteins.

Conclusions

In conclusion, it has been observed that non-specific binding is reduced when the samples under analysis are prepared in a protein matrix (human serum albumin in this case), compared to phosphate-buffered saline solution commonly used in biological research. The better results obtained in other studies in which plasma samples were analysed are due to its complex matrix which, in addition to albumin, also contains a number of other proteins, which in turn have the ability to further limit non-specific binding. The present study allows to draw, also, the following conclusions important in protein binding studies: greater attention must be paid to the processing and preparation of samples in protein matrices and to the experimental conditions, in particular temperature and pH, which must correspond to the physiological conditions in order to obtain results with biological relevance; the differences in pH, temperature, matrix composition, in obtaining standard solutions and protocols which do not ensure that the binding equilibrium is reached, especially in the case of substances with a high degree of binding, may greatly influence the experimental results. However, the *in vitro* protein binding studies by using simplified matrices represent an important step in new drug candidates' characterisation.

Conflict of interest

The authors declare no conflict of interest.

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