

## THE VALIDATION OF THE UV SPECTROPHOTOMETRIC METHOD FOR THE ASSAY OF 5 FLUOROURACIL

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### Abstract

In order to determine the release profile of 5 – fluorouracil (5-FU) from modern pharmaceutical dosage forms with vaginal administration during *in vitro* dissolution tests, an UV spectrophotometric assay method was developed, in acetate buffer solution (pH = 4.3). The working conditions were established with an absorbance maximum at  $\lambda = 266$  nm. The method was validated by determining the following parameters: the linearity in the chosen concentration domain (1.02 to 20.40  $\mu\text{g/mL}$ ), the correlation coefficient  $r^2 = 0.9992$ , the detection limit (LD) was 0.59  $\mu\text{g/mL}$ , the quantification limit (LQ) was 1.79  $\mu\text{g/mL}$ , the precision of the method (RSD = 0.5176 %), the intermediate precision of the method (RSD = 0.5521 %) and the accuracy of the method for which we obtained a mean recovery of 99.95%. In conclusion, this assay method for 5 - FU is simple, easy to apply, sensitive, linear, precise and accurate.

### Rezumat

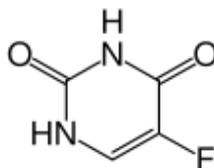
Pentru evaluarea profilului de cedare a 5 – fluorouracilului (5-FU) din forme farmaceutice moderne cu administrare vaginală, în cadrul testelor de dizolvare *in vitro*, s-a dezvoltat o metodă de dozare spectrofotometrică în UV, în soluție tampon acetat (pH = 4,3). Au fost stabilite condițiile de lucru, cu un maxim de absorbție la  $\lambda = 266$  nm. Metoda a fost validată prin determinarea următorilor parametri: liniaritatea în domeniul de concentrație ales (1,02 – 20,40  $\mu\text{g/mL}$ ), coeficientul de corelație  $r^2 = 0,9992$ , limita de detecție (LD) este de 0,59  $\mu\text{g/mL}$ , limita de cuantificare (LQ) este de 1,79  $\mu\text{g/mL}$ , precizia metodei (RSD = 0,5176 %), precizia intermediară a metodei (RSD = 0,5521 %) și acuratețea metodei pentru care s-a obținut o valoare medie a regăsirii de 99,95 %. În concluzie, această metodă de dozare a 5 – FU este simplă, ușor de aplicat, sensibilă, liniară, precisă și exactă

**Keywords:** 5 – fluorouracil, UV spectrophotometry, validation.

### Introduction

5-Fluorouracil (5-FU) is a next-generation antiviral drug, administered as vaginal topical treatment of vaginal lesions associated with the *Human papilloma virus* [3,7], including *Condylomata acuminata* [5] and

cervical intraepithelial neoplasia [7]. From the chemical point of view, 5 - FU is a fluoropyrimidine derivative, namely 5-fluoro-1 H-pyrimidine-2, 4-dione (Fig. 1).



**Figure 1.**

The chemical structure of 5-FU

In the cervical-vaginal disease therapy, 5-FU is used in several classical pharmaceutical dosage forms such as gels, creams, suppositories, as well as in some modern pharmaceutical dosage forms such as liposomes, niozomes, nanoparticles, microemulsions and vaginal films [1, 4, 6, 8, 9, 11]. Several analytical methods have been previously reported for the determination of 5 - FU, including assays based on gas liquid chromatography, and reversed-phase high-performance liquid chromatography [2,10]. The objective of this study was to develop and validate a new method for the quantitative determination of 5 - FU by UV spectrophotometry. This method will be used to assess the *in vitro* release profile of 5 - FU in various modern pharmaceutical dosage forms with vaginal administration that are under research in our department.

## Materials and Methods

### Materials

5 - Fluorouracil purity > 99% (Sigma Aldrich, Germany), sodium acetate and glacial acetic acid (Merck, Germany) were used.

**Solutions:** The acetate buffer solution (pH = 4.3) was obtained by dissolving a quantity of 1.99 g CH<sub>3</sub>COONa·3H<sub>2</sub>O and 17 mL CH<sub>3</sub>COOH 2N in a volumetric flask of 1,000 mL; the standard solution 5 - FU in acetate buffer (pH = 4.3) was obtained by dissolving a quantity of 0.1020 g of 5 - FU (reference substance) in a volumetric flask of 100 mL; the stock solution of 5 - FU was obtained by diluting 5 mL acetate buffer standard solution (pH = 4.3) in a 50 mL volumetric flask.

### Methods

*The principle of the method:* a stock solution of 5 - FU in acetate buffer solution pH = 4.3 was subjected to the UV spectrophotometric analysis at a wavelength of 266 nm in a 1 cm cell versus a blank solution consisting of acetate buffer pH = 4.3.

### Validation of the method

**Establishing the wavelength:** a sample was prepared as previously described and was submitted to the spectrophotometric analysis, then the absorption spectrum between 190-400 nm was recorded, and compared with the acetate buffer pH=4.3 in a 1 cm cell. A peak was noticed at  $\lambda = 266\text{nm}$ .

**Linearity:** Three sets of solutions containing 5 - FU were prepared in the concentration range of 1.02 - 20.40 mg/mL. For each set we separately measured the absorbance at  $\lambda = 266\text{ nm}$  and calculated the mean value. The obtained data were analyzed by linear regression (the calibration curve equation for this concentration range, the correlation coefficient ( $r$ ), the regression coefficient ( $r^2$ ), the standard error (SE) were calculated) and afterwards the calibration curve was drawn.

The limits of detection (LD) and quantification (LQ) were calculated using the following equations:

$$\text{LD} = 3.3 \times \text{standard error} / \text{slope} \quad \text{Eq. 1}$$

$$\text{LQ} = 10 \times \text{standard error} / \text{slope} \quad \text{Eq. 2}$$

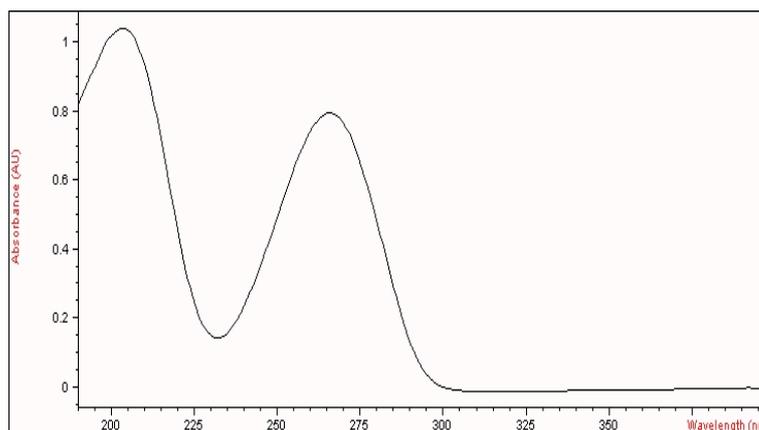
**Method precision:** we used the stock solution of 5 - FU to prepare three solutions with the following three concentrations: 7.14, 10.20 and 13.26  $\mu\text{g/mL}$  5 - FU. For each level of concentration, three samples were prepared and analyzed, and the absorbances of the obtained samples were measured experimentally.

**Intermediate precision:** the entire experiment was repeated on the following day; different reagents were used, with identical solutions and procedures as for the determination of the accuracy of the method. For each concentration, three samples were prepared and analyzed.

**Method accuracy** was assessed by the addition method: the stock solution of 5 - FU was used for preparing three solutions with the following three concentrations: 7.14, 10.20 and 13.26  $\mu\text{g/mL}$  5 - FU. The obtained solutions were analyzed spectrophotometrically following the described method. For each concentration, three samples were prepared and analyzed. Using the equation of the calibration line and the values of the corresponding absorbances, we calculated the concentration (expressed in  $\mu\text{g/mL}$ ) for each sample separately. We determined the recovery, calculated as a percentage of the value of the theoretical concentration, mean recovery, as well as the domain where recovery varies.

### Results and Discussion

Figure 2 presents the absorption spectrum for 5 - FU. Two peaks of absorption can be seen at  $\lambda = 211$  and  $\lambda = 266\text{ nm}$ . For subsequent determinations we used the absorption peak at  $\lambda = 266\text{ nm}$ .



**Figure 2.**

Absorption spectrum of 5-fluorouracil in acetate buffer (pH = 4.3)

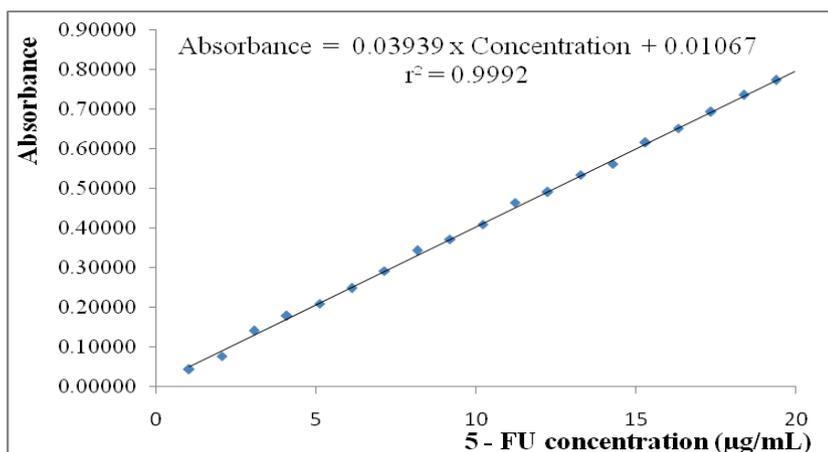
The statistical analysis of the data obtained in the determination of the linearity of the method is presented in Table I. The calibration curve and equation curve are presented in figure 3.

**Table I.**  
Method linearity

| 5-FU<br>µg/mL | Absorbance |         |         |         |
|---------------|------------|---------|---------|---------|
|               | I          | II      | III     | Average |
| 1.02          | 0.04269    | 0.04312 | 0.04289 | 0.04290 |
| 2.04          | 0.07435    | 0.07393 | 0.08392 | 0.07740 |
| 3.06          | 0.14104    | 0.14048 | 0.14049 | 0.14067 |
| 4.08          | 0.17842    | 0.17862 | 0.17822 | 0.17842 |
| 5.10          | 0.20934    | 0.20954 | 0.20890 | 0.20926 |
| 6.12          | 0.24849    | 0.24868 | 0.24953 | 0.24890 |
| 7.14          | 0.29347    | 0.29274 | 0.29192 | 0.29271 |
| 8.16          | 0.34316    | 0.34373 | 0.34376 | 0.34355 |
| 9.18          | 0.37207    | 0.37244 | 0.37299 | 0.37250 |
| 10.20         | 0.40970    | 0.41000 | 0.40919 | 0.40963 |
| 11.22         | 0.46503    | 0.46583 | 0.46549 | 0.46545 |
| 12.24         | 0.49144    | 0.49177 | 0.49057 | 0.49126 |
| 13.26         | 0.53500    | 0.53446 | 0.53440 | 0.53462 |
| 14.28         | 0.56308    | 0.56307 | 0.56198 | 0.56271 |
| 15.30         | 0.61724    | 0.61726 | 0.61608 | 0.61686 |
| 16.32         | 0.65249    | 0.65308 | 0.65238 | 0.65265 |
| 17.34         | 0.69345    | 0.69414 | 0.69369 | 0.69376 |
| 18.36         | 0.73607    | 0.73586 | 0.73634 | 0.73609 |
| 19.38         | 0.77445    | 0.77499 | 0.77406 | 0.77450 |
| 20.40         | 0.80580    | 0.80648 | 0.80668 | 0.80632 |

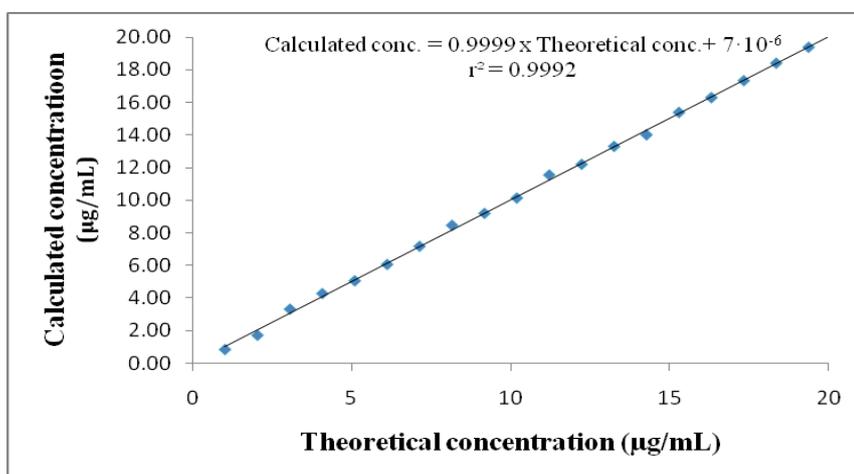
**$r = 0.996$ ;  $r^2 = 0.9992$ ; SE = 0.0070; Intercept = 0.01067; Slope = 0.0393**

SE – standard error, r – correlation coefficient,  $r^2$  – regression coefficient

**Figure 3.**

The calibration curve and equation calibration for determining 5-FU through UV spectrophotometry

Estimating the linearity of the results as seen in figure 4, there is a linear correlation between the theoretically introduced concentration and the calculated concentration, using the calibration curve equation, the slope of this being very close to the unity (slope = 0.9999) and the intercept having the value of  $7 \times 10^{-6}$ . The correlation coefficient has the value  $r^2 = 0.9992$ . In conclusion, in the considered concentration domain we considered, between the values of the calculated and the theoretical concentrations, there is a very good correlation.

**Figure 4.**

Linearity results for the assay of 5-FU through UV spectrophotometry

The values of the detection and quantification limits are the following: LD = 0.59  $\mu\text{g/mL}$  and LQ = 1.79  $\mu\text{g/mL}$ .

The experimental values obtained in the study of precision and intermediate precision of the assay method of 5 - FU through UV spectrophotometry are presented in Table II.

**Table II.**  
Precision and accuracy of the assay method of 5-FU through UV spectrophotometry

| 5 - FU concentration ( $\mu\text{g/mL}$ ) | Method precision                                |              | Intermediate precision                          |              | Method accuracy                                       |              |
|---|---|--------------|---|--------------|---|--------------|
|   | Absorbance                                      | Recovery (%) | Absorbance                                      | Recovery (%) | Absorbance  | Recovery (%) |
| 7.14                                      | 0.2934  | 100.56       | 0.2924  | 100.14       | 0.2932  | 100.42       |
|   | 0.2927  | 100.28       | 0.2928  | 100.28       | 0.2928  | 100.28       |
|   | 0.2919  | 100.00       | 0.2930  | 100.42       | 0.2920  | 100.00       |
|   | 0.4097  | 99.31        | 0.4099  | 99.41        | 0.4096  | 99.31        |
| 10.20                                     | 0.4100  | 99.41        | 0.4096  | 99.31        | 0.4099  | 99.41        |
|   | 0.4091  | 99.22        | 0.4084  | 99.02        | 0.4093  | 99.22        |
|   | 0.5350  | 100.38       | 0.5341  | 100.23       | 0.5340  | 100.23       |
| 13.26                                     | 0.5344  | 100.30       | 0.5349  | 100.38       | 0.5346  | 100.30       |
|   | 0.5344  | 100.30       | 0.5352  | 100.45       | 0.5351  | 100.38       |
| Statistical data                          | Mean = 99.97 %<br>SD = 0.5175<br>RSD = 0.5176 % |              | Mean = 99.96 %<br>SD = 0.5519<br>RSD = 0.5521 % |              | Mean = 99.95 %<br>Minim = 99.22 %<br>Maxim = 100.42 % |              |

SD – standard deviation, RSD – relative standard deviation

According to the experimental data and after their statistical processing, it is clear that on a range of 30% compared to the value of interest (10  $\mu\text{g/mL}$ ), the method is precise, as the relative standard deviation (RSD) is below the required limit (5%).

### Conclusions

In this study we presented the results obtained in the development and validation of an assay method of 5 - FU in sodium acetate buffer solution (pH = 4.3) by UV spectrophotometry. The method was validated by determining the following parameters: linearity in the chosen concentration interval (1.02 - 20.40  $\mu\text{g/mL}$ ),  $r = 0.9996$ ,  $r^2 = 0.9992$ , LD = 0.59  $\mu\text{g/mL}$ , LQ = 1.79  $\mu\text{g/mL}$ , precision of the method (RSD = 0.5176%), intermediate precision of the method (RSD = 0.5521%) method accuracy for which we obtained a mean recovery of 99.95 %.

In conclusion, this method for the assay of 5 - FU is simple, easy to apply, sensitive, linear, precise and accurate, and can be used in future research for the quantitative evaluation of 5 - FU in modern pharmaceutical dosage forms with vaginal administration during *in vitro* dissolution tests.

#### Acknowledgments

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